

* * * * * Welcome to STN International * * * * *

<u>NEWS</u>	<u>1</u>	Web Page URLs for STN Seminar Schedule - N. America
<u>NEWS</u>	<u>2</u>	"Ask CAS" for self-help around the clock
<u>NEWS</u>	<u>3</u>	May 12 EXTEND option available in structure searching
<u>NEWS</u>	<u>4</u>	May 12 Polymer links for the POLYLINK command completed in REGISTRY
<u>NEWS</u>	<u>5</u>	May 27 New UPM (Update Code Maximum) field for more efficient patent SDIs in CAplus
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<u>NEWS</u>	<u>7</u>	Jun 28 Additional enzyme-catalyzed reactions added to CASREACT
<u>NEWS</u>	<u>8</u>	Jun 28 ANTE, AQUALINE, BIOENG, CIVILENG, ENVIROENG, MECHENG, and WATER from CSA now available on STN(R)
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<u>NEWS</u>	<u>11</u>	AUG 02 IFIPAT/IFIUDB/IFICDB reloaded with new search and display fields
<u>NEWS</u>	<u>12</u>	AUG 02 CAplus and CA patent records enhanced with European and Japan Patent Office Classifications
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<u>NEWS</u>	<u>14</u>	AUG 02 The Analysis Edition of STN Express with Discover! (Version 7.01 for Windows) now available
<u>NEWS</u>	<u>15</u>	AUG 04 Pricing for the Save Answers for SciFinder Wizard within STN Express with Discover! will change September 1, 2004
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NEWS EXPRESS JULY 30 CURRENT WINDOWS VERSION IS V7.01, CURRENT MACINTOSH VERSION IS V6.0c(ENG) AND V6.0Jc(JP), AND CURRENT DISCOVER FILE IS DATED 11 AUGUST 2004

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FILE COVERS 1907 - 14 Sep 2004 VOL 141 ISS 12
 FILE LAST UPDATED: 13 Sep 2004 (20040913/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

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=> s (ecological or smokeless) (P) coal
  9916 ECOLOGICAL
    4 ECOLOGICALS
  9918 ECOLOGICAL
    (ECOLOGICAL OR ECOLOGICALS)
  34118 ECOL
    1 ECOLS
  34119 ECOL
    (ECOL OR ECOLS)
  38406 ECOLOGICAL
    (ECOLOGICAL OR ECOL)
  2552 SMOKELESS
  209512 COAL
  35522 COALS
  211340 COAL
    (COAL OR COALS)
L1      1238 (ECOLOGICAL OR SMOKELESS) (P) COAL
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```
=> s l1 and carbonate
  252133 CARBONATE
  62405 CARBONATES
  282963 CARBONATE
    (CARBONATE OR CARBONATES)
L2      19 L1 AND CARBONATE
```

=> d l2 1-19 all

L2 ANSWER 1 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

<input checked="" type="checkbox"/> Full Text	<input type="checkbox"/> Citing References
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AN 2004:191307 CAPLUS
 DN 141:9413
 ED Entered STN: 10 Mar 2004
 TI Basic physicochemical principles of heat treatment of sulfur-containing wastes from coal mining and beneficiation
 AU Shpirt, M. Ya.; Goryunova, N. P.; Zil'bershmidt, M. G.; Samuilov, E. V.; Goryunova, E. V.
 CS Inst. Goryuchikh Iskopayemykh, Russia
 SO Khimiya Tverdogo Topliva (Moscow, Russian Federation) (2004), (1), 64-80
 CODEN: KTVTBY; ISSN: 0023-1177
 PB Nauka
 DT Journal
 LA Russian
 CC 51-19 (Fossil Fuels, Derivatives, and Related Products)

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- Section cross-reference(s): 59, 60, 61
- AB It is shown (for the example of mining waste and waste from beneficiation of brown coal from the Podmoskov Basin) that storage and disposal of sulfur-contg. coal mining waste (enrichment) may be accompanied by unfavorable effects on the surrounding environment (increases levels of toxic oxides and ecol. dangerous substances in the atm. and groundwater-sulfuric acid, aluminum compds., iron, arsenic, manganese, chromium, zinc, nickel, and others). Thermodn. modeling and exptl. investigations on the heat treatment of waste streams without additives were studied and the results are presented. It is shown that certain conditions with more than 90% sulfur, contained in the model, can be transferred in the gas phase from appearing a mixt. of SO₂ and SO₃, but in the solid state can be received in iron compds., which can be sepd. and removed magnetically. Compns. and elemental distribution between phases depends upon both the temp. of reaction and the oxygen level in the gas/air feed. Implications and ways of dealing with the technol. dangers of the processes, practical characteristics of the liq. and solid streams are also given, in terms of concns. of sulfuric acid, iron (>50% in the form of Fe₂O₃), aluminum sulfate and sulfur.
- ST heat treatment sulfur waste coal mining beneficiation pollution control
- IT Filtration
 Heat treatment
 Magnetic separation
 Neutralization
 Oxidation
 Thermodynamic simulation
 (basic physicochem. principles of heat treatment of sulfur-contg. wastes from coal mining and beneficiation)
- IT Sulfates, properties
 RL: FMU (Formation, unclassified); OCU (Occurrence, unclassified); PRP (Properties); FORM (Formation, nonpreparative); OCCU (Occurrence)
 (basic physicochem. principles of heat treatment of sulfur-contg. wastes from coal mining and beneficiation)
- IT Carbonates, reactions
 Oxides (inorganic), reactions
 RL: FMU (Formation, unclassified); OCU (Occurrence, unclassified); PRP (Properties); RCT (Reactant); FORM (Formation, nonpreparative); OCCU (Occurrence); RACT (Reactant or reagent)
 (basic physicochem. principles of heat treatment of sulfur-contg. wastes from coal mining and beneficiation)
- IT Sulfides, reactions
 Trace metals
 RL: OCU (Occurrence, unclassified); POL (Pollutant); RCT (Reactant); OCCU (Occurrence); RACT (Reactant or reagent)
 (basic physicochem. principles of heat treatment of sulfur-contg. wastes from coal mining and beneficiation)
- IT Coal treatment
 (cleaning, wastes; basic physicochem. principles of heat treatment of sulfur-contg. wastes from coal mining and beneficiation)
- IT Solid wastes
 (coal washing; basic physicochem. principles of heat treatment of sulfur-contg. wastes from coal mining and beneficiation)
- IT Mining
 (coal, waste from; basic physicochem. principles of heat treatment of sulfur-contg. wastes from coal mining and beneficiation)
- IT Solid wastes
 (mine, coal mines; basic physicochem. principles of heat treatment of sulfur-contg. wastes from coal mining and beneficiation)
- IT Toxicity
 (of waste materials; basic physicochem. principles of heat treatment of sulfur-contg. wastes from coal mining and beneficiation)
- IT 7429-90-5D, Aluminum, compds. 7439-96-5D, Manganese, compds.
 7440-02-0D, Nickel, compds. 7440-38-2D, Arsenic, compds. 7440-47-3D,
 Chromium, compds. 7440-66-6D, Zinc, compds.

- RL: FMU (Formation, unclassified); OCU (Occurrence, unclassified); POL (Pollutant); PRP (Properties); RCT (Reactant); FORM (Formation, nonpreparative); OCCU (Occurrence); RACT (Reactant or reagent)
 (basic physicochem. principles of heat treatment of sulfur-contg.
 wastes from coal mining and beneficiation)
- IT 1309-37-1, Ferric oxide, reactions 7439-89-6D, Iron, compds.
- RL: FMU (Formation, unclassified); OCU (Occurrence, unclassified); PRP (Properties); RCT (Reactant); FORM (Formation, nonpreparative); OCCU (Occurrence); RACT (Reactant or reagent)
 (basic physicochem. principles of heat treatment of sulfur-contg.
 wastes from coal mining and beneficiation)
- IT 10043-01-3, Aluminum sulfate
 RL: FMU (Formation, unclassified); POL (Pollutant); FORM (Formation, nonpreparative); OCCU (Occurrence)
 (basic physicochem. principles of heat treatment of sulfur-contg.
 wastes from coal mining and beneficiation)
- IT 7664-93-9, Sulfuric acid, properties
 RL: FMU (Formation, unclassified); POL (Pollutant); PRP (Properties); FORM (Formation, nonpreparative); OCCU (Occurrence)
 (basic physicochem. principles of heat treatment of sulfur-contg.
 wastes from coal mining and beneficiation)
- IT 1305-78-8, Calcium oxide, reactions 1309-48-4, Magnesium oxide,
 reactions 1344-28-1, Alumina, reactions 7631-86-9, Silica, reactions
 RL: OCU (Occurrence, unclassified); PEP (Physical, engineering or chemical process); PYP (Physical process); RCT (Reactant); OCCU (Occurrence); PROC (Process); RACT (Reactant or reagent)
 (basic physicochem. principles of heat treatment of sulfur-contg.
 wastes from coal mining and beneficiation)

L2 ANSWER 2 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

	Full Text	Citing References			
AN	2003:810150 CAPLUS				
DN	139:278831				
ED	Entered STN: 16 Oct 2003				
TI	Ecologically friendly fuel and its production				
IN	Macho, Vendelin; Bakos, Dusan; Bajus, Martin; Cicmanec, Peter; Boroska, Fedor; Zatko, Ludovit; Komora, Ladislav				
PA	Hornonitrianske Bane Prievidza, A. S., Slovakia; Chemickotechnologicka Fakulta STU				
SO	Slovakia, 7 pp.				
	CODEN: SLXXFO				
DT	Patent				
LA	Slovak				
IC	ICM C10L009-10 ICS C10L010-00; C10L005-16; B01D053-34				
CC	51-12 (Fossil Fuels, Derivatives, and Related Products) Section cross-reference(s): 52, 59				
FAN.CNT 1					
	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	<u>SK 282533</u>	B6	20021008	<u>SK 1995-870</u>	19950706
PRAI	<u>SK 1995-870</u>		19950706		

CLASS	PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
SK 282533	ICM	C10L009-10 ICS C10L010-00; C10L005-16; B01D053-34	

AB The solid, semisolid, and/or liq. fuel having a calorific value of 8-42 MJ/kg and contg. 0.1-6 wt.% free and/or bonded S and optional As is mixed with 50-800% excess Ca and/or Mg in the form of ≥1 compd. which can react with SO₂ and/or SO₃ at 300-1,200°. The Ca- and/or Mg-contg. additives are added to the fuel prior and/or during combustion. Amt. of the additives in the fuel is 0.5-30 wt.%.

ST ecol friendly fuel prodn; nonpolluting fuel prodn
 IT Lime (chemical)
 Limestone, uses
 RL: MOA (Modifier or additive use); USES (Uses)
 (additive for bonding of sulfur and arsenic in prepn. of ecol. friendly
 fuel)

IT Bark
Coal slurries
 Fuel briquets
 Fuel oil
 Peat
 Petroleum refining residues
 Sawdust
 Soot
 Straw
 (bonding of sulfur and arsenic compds. in prepn. of **ecol.**
 friendly fuel from)

IT Anthracite
 Asphalt
 Asphaltenes
 Bitumens
Brown coal
Coal, processes
 Coke
 Lignite
 Petroleum coke
 Semicoke
 Tar
 RL: CPS (Chemical process); PEP (Physical, engineering or chemical
 process); PROC (Process)
 (bonding of sulfur and arsenic compds. in prepn. of **ecol.**
 friendly fuel from)

IT Carboxylic acids, uses
 RL: MOA (Modifier or additive use); USES (Uses)
 (calcium salts; additive for bonding of sulfur and arsenic in prepn. of
 ecol. friendly fuel)

IT Air pollution
 (control; bonding of sulfur and arsenic compds. in prepn. of ecol.
 friendly fuel)

IT Petroleum products
 (fractions; bonding of sulfur and arsenic compds. in prepn. of ecol.
 friendly fuel from)

IT Surfactants
 (in bonding of sulfur and arsenic compds. in prepn. of ecol. friendly
 fuels)

IT Carboxylic acids, uses
 RL: MOA (Modifier or additive use); USES (Uses)
 (magnesium salts; additive for bonding of sulfur and arsenic in prepn.
 of ecol. friendly fuel)

IT Polymers, processes
 Rubber, processes
 RL: CPS (Chemical process); PEP (Physical, engineering or chemical
 process); PROC (Process)
 (waste; bonding of sulfur and arsenic compds. in prepn. of ecol.
 friendly fuel from)

IT 471-34-1, Calcium carbonate, uses 546-93-0, Magnesium
carbonate 1305-62-0, Calcium hydroxide, uses 1305-78-8,
 Calcium oxide, uses 1309-42-8, Magnesium hydroxide 1309-48-4,
 Magnesium oxide, uses 7000-29-5, Calcium magnesium **carbonate**
7439-95-4D, Magnesium, alcoholate or phenolate 7440-70-2D, Calcium,
 alcoholate or phenolate 13717-00-5, Magnesite 16389-88-1, Dolomite,
 uses 27576-86-9D, Cumylphenol, calcium salt
 RL: MOA (Modifier or additive use); USES (Uses)
 (additive for bonding of sulfur and arsenic in prepn. of ecol. friendly

fuel)

- IT 9016-45-9, Polyethylene glycol nonylphenyl ether
 RL: MOA (Modifier or additive use); USES (Uses)
 (surfactant in bonding of sulfur and arsenic compds. in prepn. of ecol.
 friendly fuels)
- IT 9003-07-0, Polypropylene
 RL: CPS (Chemical process); PEP (Physical, engineering or chemical
 process); PROC (Process)
 (waste; bonding of sulfur and arsenic compds. in prepn. of ecol.
 friendly fuel from)

L2 ANSWER 3 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

Full Text	Citing References
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AN 2003:467234 CAPLUS
 DN 139:105161
 ED Entered STN: 19 Jun 2003
 TI Method of production of components of cement-free hydraulic binder
 IN Shchukin, V. S.
 PA Ukraine
 SO Russ., No pp. given
 CODEN: RUXXE7
 DT Patent
 LA Russian
 IC ICM C04B007-44
 CC 58-3 (Cement, Concrete, and Related Building Materials)

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
<u>RU 2200137</u>	C2	20030310	<u>RU 2000-118835</u>	20000718
<u>PRAI UA 1999-116166</u>	A	19991111		

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
RU 2200137	ICM	C04B007-44

AB The method includes prepn. of raw material mixt. contg. SiO₂, CaO, MgO, Al₂O₃, FeO, and Fe₂O₃ followed by heat treatment, melting, cooling, and grinding of product thus obtained. Content of said oxides in the raw material mixt. is ≥70%; heat treatment is performed in presence of reducing agent by heating to 1300–500°, after which the melt is heated to 1800° and iron-contg. product is sepd.; then, the melt is cooled, and grinding of the product is performed after granulation of the melt; oxides contained in the granulate have the following ratios: (CaO+MgO)/SiO₂ = 0.5–1.7, and Al₂O₃/SiO₂ = 0.3–0.6. Water in the amt. of 30–35 wt.% is added to ground product. Granulation of the melt is performed by cooling it with natural gas; gas formed after granulation of the melt is used as reductant. Wastes of coal processing or low-grade coal may also be used as reducing agents. Ground product may be addnl. mixed with finely-dispersed filler at ratio of 1 : (0.1–2). Ash-and-slag wastes of thermal power plants, coal fly ash from thermal power plants and red mud of silica prodn. may be used as finely-dispersed filler. Solns. of NaOH, KOH, Na₂CO₃, Na₂SO₃, K₂SO₃, and/or CaSO₃ may be added to mixt. in the amt. of 2–10 wt.%. The method provides decrease of energy consumption, improved quality of the product, and enhanced ecol. safety.

- ST fly ash coal lime waste recycling
 IT Red mud (bauxite processing residue)
 (additive of cement-free binder; method of prodn. of components of cement-free hydraulic binder)
 IT Recycling
 (ash and slag wastes; method of prodn. of components of cement-free hydraulic binder)
 IT Ashes (residues)
 (coal fly; method of prodn. of components of cement-free hydraulic binder)

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IT Lime (chemical)
 RL: TEM (Technical or engineered material use); USES (Uses)
 (hydraulic binder component; method of prodn. of components of cement-free hydraulic binder)

IT Binders
 (hydraulic, cement-free; method of prodn. of components of cement-free hydraulic binder)

IT 497-19-8, Sodium **carbonate** (Na₂CO₃), uses 1310-58-3, Potassium hydroxide (KOH), uses 1310-73-2, Sodium hydroxide (NaOH), uses 7757-83-7, Sodium sulfite (Na₂SO₃) 10117-38-1, Potassium sulfite (K₂SO₃) 10257-55-3, Calcium sulfite (CaSO₃)
 RL: TEM (Technical or engineered material use); USES (Uses)
 (additive of cement-free binder; method of prodn. of components of cement-free hydraulic binder)

L2 ANSWER 4 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

Full Text	Citing References
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AN 2003:431338 CAPLUS
 DN 139:296259
 ED Entered STN: 05 Jun 2003
 TI Comparison of Asian clam field bioassays and benthic community surveys in quantifying effects of a coal-fired power plant effluent on Clinch River biota
 AU Hull, M. S.; Cherry, D. S.; Soucek, D. J.; Currie, R. J.; Neves, R. J.
 CS Department of Biology, Virginia Polytechnic Institute and State University, Blacksburg, VA, USA
 SO Journal of Aquatic Ecosystem Stress and Recovery (2002), 9(4), 271-283
 CODEN: JASRF9; ISSN: 1386-1980
 PB Kluwer Academic Publishers
 DT Journal
 LA English
 CC 61-2 (Water)
 Section cross-reference(s): 12, 51, 60
 AB Asian clam Survival and growth may be more sensitive endpoints than benthic macroinvertebrate community richness parameters at distinguishing biotic impairment attributable to complex effluent from **coal**-burning utilities. Field bioassays were performed with the Asian clam, Corbicula fluminea, 2000-2002, and rapid bioassessments of benthic macroinvertebrate communities, 2000-2001, at sites upstream and downstream of the American Elec. Power (AEP) Clinch River Plant (CRP) in Russell County, Virginia. Survival and growth of transplanted C. fluminea were significantly impaired within the CRP effluent plume (avs. of 35% and 0.21 mm, resp.) relative to all other study sites within the Clinch River (avs. of 89% and 1.58 mm). Conversely, richness metrics for Ephemeroptera, Ephemeroptera-Plecoptera-Trichoptera (EPT), and total taxa were not reduced downstream from the CRP; however, relative abundance metrics for Ephemeroptera and EPT were minimally reduced at the CRP-affected site in 2000-01. Results suggested that richness metrics for benthic macroinvertebrate communities may be inadequate to assess the effect of complex industrial wastewater on C. fluminea. Results have implications for bioassessment methods used to monitor streams inhabited by imperiled freshwater mussels because C. fluminea and Unionoidea are **ecol.** similar and recent findings suggested certain Unionidae genera may be more sensitive than C. fluminea.
 ST water pollution coal fired power effluent Clinch River Virginia; Asian clam field bioassay river pollution power generation effluent; benthic macroinvertebrate community survey river pollution power generation effluent; survival growth Asain clam polluted water Clinch River
 IT Alkalinity
 Benthic organisms
 Corbicula fluminea
 Electric conductivity
 Ephemeroptera

Plecoptera
 Temperature
 Trichoptera
 (Asian clam field bioassays vs. benthic macroinvertebrate community surveys to quantify coal-fired power generation effluent effect on aquatic biota, Clinch River, Virginia)

IT Bioassay
 (Asian clam field; Asian clam field bioassays vs. benthic macroinvertebrate community surveys to quantify coal-fired power generation effluent effect on aquatic biota, Clinch River, Virginia)

IT Growth, animal
 (Asian clam survival and; Asian clam field bioassays vs. benthic macroinvertebrate community surveys to quantify coal-fired power generation effluent effect on aquatic biota, Clinch River, Virginia)

IT Biota (**ecological** unit)
 (biodiversity, unionoid; Asian clam field bioassays vs. benthic macroinvertebrate community surveys to quantify **coal**-fired power generation effluent effect on aquatic biota, Clinch River, Virginia)

IT Power
 (coal-fired generation of; Asian clam field bioassays vs. benthic macroinvertebrate community surveys to quantify coal-fired power generation effluent effect on aquatic biota, Clinch River, Virginia)

IT Landfill leachate
 (coal-fired power generation fly ash; Asian clam field bioassays vs. benthic macroinvertebrate community surveys to quantify coal-fired power generation effluent effect on aquatic biota, Clinch River, Virginia)

IT Toxicity
 (power generation effluent; Asian clam field bioassays vs. benthic macroinvertebrate community surveys to quantify coal-fired power generation effluent effect on aquatic biota, Clinch River, Virginia)

IT Coal, uses
 RL: NUU (Other use, unclassified); USES (Uses)
 (power generation from combustion of; Asian clam field bioassays vs. benthic macroinvertebrate community surveys to quantify coal-fired power generation effluent effect on aquatic biota, Clinch River, Virginia)

IT Water pollution
 (river water; Asian clam field bioassays vs. benthic macroinvertebrate community surveys to quantify coal-fired power generation effluent effect on aquatic biota, Clinch River, Virginia)

IT 7782-44-7, Oxygen, occurrence 12408-02-5, Hydrogen ion, occurrence
 RL: OCU (Occurrence, unclassified); OCCU (Occurrence)
 (Asian clam field bioassays vs. benthic macroinvertebrate community surveys to quantify coal-fired power generation effluent effect on aquatic biota, Clinch River, Virginia)

IT 471-34-1, Calcium **carbonate**, occurrence
 RL: OCU (Occurrence, unclassified); OCCU (Occurrence)
 (hardness as; Asian clam field bioassays vs. benthic macroinvertebrate community surveys to quantify coal-fired power generation effluent effect on aquatic biota, Clinch River, Virginia)

RE.CNT 65 THERE ARE 65 CITED REFERENCES AVAILABLE FOR THIS RECORD

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L2 ANSWER 5 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

Full Text Citing References

AN 2002:878666 CAPLUS
 DN 138:109815
 ED Entered STN: 20 Nov 2002
 TI New data on the geology, mineral resources and geo-ecology of Franz-Josef Land Archipelago
 AU Makar'ev, A. A.; Makar'eva, E. M.; Kosteva, N. N.
 CS Russia
 SO Razvedka i Okhrana Nedr (2002), (9), 23-27
 CODEN: RZONAV; ISSN: 0034-026X
 PB Nedra
 DT Journal
 LA Russian
 CC 53-12 (Mineralogical and Geological Chemistry)
 Section cross-reference(s): 19, 51
 AB Geol., islands of the Franz-Josef Land Archipelago have a basement that includes folded Vendian rocks and Carboniferous **coal**-bearing clastic and **carbonate** rocks, and is overlain by Mesozoic sedimentary rocks of a platform cover. Trap formations are widespread: dikes, sills, and stocks of gabbro and gabbro-diorite; necks of microdolerites and hyalobasalts; basalt and andesitic-basalt lavas, and their tuffs and tuff-lavas. In chem. compn. all of the igneous rocks belong to basic normal series with sodic and sodic-potassic alkyl. In comparison to the clarke values for basic igneous rocks, the av. contents of Ga, Sn, Mo, V, Cu, Zn, Zr, and Ce are higher and those of Cr, Nb, Ag, Sr, and Sc are lower. The known mineral resources include bituminous **coal**, bituminous rocks, siallite, and stone of com. value. At storage sites of fuels and lubricants the soils are polluted by petroleum products. Also, soils and moss have ^{137}Cs \leq 352 and \leq 350 Bq/kg, resp., and ^{60}Co \leq 60 and \leq 115 Bq/kg, resp. In the waters adjacent to the archipelago the Quaternary sediments are no more than 3-5 m thick, and the bottom sediments are **ecol.** undisturbed.
 ST geol mineral resource geoecol Franz Josef Land Archipelago
 IT Basalt
 RL: GOC (Geological or astronomical occurrence); PRP (Properties); OCCU (Occurrence)
 (andesitic; geol., mineral resources and geo-ecol. of Franz-Josef Land Archipelago, Russia)
 IT Dikes
 (basic; geol., mineral resources and geo-ecol. of Franz-Josef Land Archipelago, Russia)
 IT Moss
 Soil pollution
 (cobalt-60 and cesium-137 in moss and soil, Franz-Josef Land Archipelago, Russia)
 IT Trace elements, occurrence
 RL: GOC (Geological or astronomical occurrence); GPR (Geological or astronomical process); OCCU (Occurrence); PROC (Process)
 (geol. indicator, igneous rocks; in basic igneous rocks, of Franz-Josef Land Archipelago, Russia)
 IT Basic magmatism
 Geological sediments
 (geol., mineral resources and geo-ecol. of Franz-Josef Land Archipelago, Russia)
 IT **Coal**, occurrence
 Nonmetal ores
 Siallite

h ebc g cg b cg

eb

Tuff
 RL: GOC (Geological or astronomical occurrence); OCCU (Occurrence)
 (geol., mineral resources and geo-**ecol.** of Franz-Josef Land
 Archipelago, Russia)

IT Andesite
 Basic igneous rocks
 Diorite
 Dolerite
 Gabbro
 Sedimentary rocks
 RL: GOC (Geological or astronomical occurrence); PRP (Properties); OCCU
 (Occurrence)
 (geol., mineral resources and geo-ecol. of Franz-Josef Land
 Archipelago, Russia)

IT Lava
 RL: GOC (Geological or astronomical occurrence); PRP (Properties); OCCU
 (Occurrence)
 (mafic; geol., mineral resources and geo-ecol. of Franz-Josef Land
 Archipelago, Russia)

IT Basic igneous rocks
 Ultrabasic igneous rocks
 RL: GOC (Geological or astronomical occurrence); PRP (Properties); OCCU
 (Occurrence)
 (trap rock; geol., mineral resources and geo-ecol. of Franz-Josef Land
 Archipelago, Russia)

IT 10045-97-3, Cesium-137, occurrence 10198-40-0, Cobalt 60, occurrence
 RL: POL (Pollutant); OCCU (Occurrence)
 (in moss and soils; geol., mineral resources and geo-ecol. of
 Franz-Josef Land Archipelago, Russia)

L2 ANSWER 6 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

	Full Text	Citing References
AN	2001:79098	CAPLUS
DN	134:256523	
ED	Entered STN: 04 Feb 2001	
TI	Development of environmentally benign scale inhibitors for industrial applications	
AU	Hater, Wolfgang; Mayer, Bernd; Schweinsberg, Matthias	
CS	Germany	
SO	PowerPlant Chemistry (2000), 2(12), 721-724, 752-755	
	CODEN: POCHFT; ISSN: 1438-5325	
PB	PowerPlant Chemistry GmbH	
DT	Journal	
LA	English	
CC	61-6 (Water)	
AB	Polyaspartic acid and polysaccharide derivs. were used as starting materials for the development of an ecol. sound scale inhibitor. BaSO ₄ , CaSO ₄ , and CaCO ₃ stabilization was tested and the results were compared with those of products based on phosphonic acids. Of all the inhibitors tested, only polyaspartates exhibit good scale inhibition against all 3 minerals, whereas phosphonates are completely ineffective against CaSO ₄ and saccharides exhibit inferior inhibition against BaSO ₄ scale. Two field tests on the application of inhibitors on the base of polyaspartates are described: BaSO ₄ inhibition in coal mine drainage and CaSO ₄ inhibition at a power station.	
ST	water scale inhibitor polyaspartate polysaccharide	
IT	Environmental pollution control	
	Scale inhibitors (environmentally benign scale inhibitors for industrial applications)	
IT	Polysaccharides, processes	
	RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process) (environmentally benign scale inhibitors for industrial applications)	

contg.)
IT 471-34-1, Calcium **carbonate**, processes 7727-43-7, Barium sulfate 7778-18-9, Calcium sulfate 25608-40-6, Polyaspartic acid
RL: PEP (Physical, engineering or chemical process); PRP (Properties);
PROC (Process)
(environnally benign scale inhibitors for industrial applications
contg.)

RE.CNT 12 THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS RECORD

RE

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- (12) Steber, J; Chemosphere 1968, V15(7), P929

L2 ANSWER 7 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

Full Text	Citing References
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AN 2000:660620 CAPLUS
DN 133:226892
ED Entered STN: 21 Sep 2000
TI Influence of addition of alkali metal compounds to calcium **carbonate** on desulfurization characteristics
AU Naruse, Ichiro; Saito, Katsuhiro; Murakami, Takahiro
CS Department of Ecological Engineering, Toyohashi University of Technology, Tempaku-cho, Toyohashi, 441-8580, Japan
SO Proceedings of the ASME/JSME Thermal Engineering Joint Conference, 5th, San Diego, CA, United States, Mar. 14-19, 1999 (1999), 1347-1353
Publisher: American Society of Mechanical Engineers, New York, N. Y.
CODEN: 69AEKP
DT Conference; (computer optical disk)
LA English
CC 59-4 (Air Pollution and Industrial Hygiene)
AB Acid rain has been recently involved as one of serious global environmental problems, esp. sulfur oxide (SOx) has been already recognized to be one of the greatest sources. Limestone is currently supplied as a desulfurizer into bubbling and circulating fluidized bed **coal** combustors since both combustors are operated at the temp. ranged from 1073 to 1173 K, where limestone can be calcined and sulfurized optimally. In the practical boilers, however, the limestone particles are fed to the combustor excessively since the utilization efficiency of CaO produced by the calcination of limestone is low. This phenomenon is caused by the plugging of pores due to CaSO4 formation. Thereafter, this operation causes the increase of ash vol. and then CO2 concn. in the atm. On the other hand, many kinds of sea-shell are clarified as one of industrial wastes, and also consist of CaCO3 similar to limestone. Therefore it would be possible for wasted sea-shell to be applied to one of the desulfurizers. In this case the CO2 produced by calcination of the shell is fixed and recycled naturally in obedience to the **ecol.** law. From this viewpoint, desulfurization characteristics of wasted sea shell have been already studied fundamentally by using a thermobalance as compared with the results obtained by limestone. As one of the main results in this study, the desulfurizaton efficiency for the shell attained a value of more than about 70%. For the limestone, on the other hand, it is less than 40%. It was also explained that the obtained result was caused by the difference of the pore size distribution of CaO between

limestone and sea-shell. In order to elucidate the reason of this big difference between the sea-shell and limestone, alkali metal compds. contained in the shell were removed by extg. in deionized water as well as chloride or **carbonate** compds. of alkali metals were phys. mixed with the limestone in this study. Both the extd. shell particles and the mixt. of limestone with alkali metal compds. were also tested by using the thermobalance to discuss the detailed role of alkali metal compds. in the sea-shell. In the case that alkali metal compds. are added to limestone, on the other hand, particles of the alkali compd. are phys. mixed with particles of limestone. NaCl, KCl, Na₂CO₃ and K₂CO₃ were used as the alkali metal compds. to compare the effect of chloride compd. with that of **carbonate** compd. The results obtained by this study are summarized as follows. I) The desulfurization activity for wasted sea-shell is much higher than that for limestone. II) Even if the alkali metal compds. are partially removed from the sea shell, the desulfurization efficiency does not change. III) The desulfurization activity can be enhanced by adding alkali metal compds. to limestone. Sodium compds. are more effective on the desulfurization efficiency than potassium compds. Sodium chloride is the best agent among them.

ST flue gas desulfurization seashell calcium **carbonate**

IT Flue gas desulfurization

Shell

(influence of addn. of alkali metal compds. to seashell-derived calcium **carbonate** on flue gas desulfurization characteristics)

IT 497-19-8, Sodium **carbonate**, uses 584-08-7, Potassium **carbonate** 7447-40-7, Potassium chloride, uses 7647-14-5, Sodium chloride, uses

RL: MOA (Modifier or additive use); USES (Uses)

(influence of addn. of alkali metal compds. to seashell-derived calcium **carbonate** on flue gas desulfurization characteristics)

IT 471-34-1, Calcium **carbonate**, uses

RL: NUU (Other use, unclassified); USES (Uses)

(influence of addn. of alkali metal compds. to seashell-derived calcium **carbonate** on flue gas desulfurization characteristics)

IT 7446-09-5, Sulfur dioxide, processes

RL: REM (Removal or disposal); PROC (Process)

(influence of addn. of alkali metal compds. to seashell-derived calcium **carbonate** on flue gas desulfurization characteristics)

RE.CNT 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD

RE

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- (2) Hartman, M; Ind Eng Chem Process Des Dev 1974, V13(3), P248 CAPLUS
- (3) Naruse, I; Kagaku Kogaku Ronbunshu 1995, V21(5), P904 CAPLUS

L2 ANSWER 8 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

Full	Citing
Text	References

AN 1999:219338 CAPLUS
 DN 130:286582
 ED Entered STN: 08 Apr 1999
 TI Ecological impact produced by hard **coal** mining in Petrosani **coal** basin and pollution reduction at Jiu river
 AU Arad, V.; Arad, S.; Marchis, Gh.
 CS Petrosani University, Rom.
 SO Environmental Issues and Waste Management in Energy and Mineral Production, Proceedings of the International Symposium on Environmental Issues and Waste Management in Energy and Mineral Production, 5th, Ankara, May 18-20, 1998 (1998), 319-322. Editor(s): Pasamehmetoglu, A. Gunhan; Ozgenoglu, Abdurrahim. Publisher: Balkema, Rotterdam, Neth.
 CODEN: 67KRAM
 DT Conference
 LA English
 CC 61-2 (Water)
 Section cross-reference(s): 51, 60

AB Economical development of the Petrosani coal basin (Romania) based on hard coal unit exploitation and processing has a dynamic character without accounting for environmental aspects. Major environmental problems consist of pollution of Jiu River as a consequence of the effluent rejected in water. Aspects of environmental impact induced by the hard coal mining Petrosani coal basin are discussed.

ST coal mining processing water pollution Petrosani Romania

IT Mining
 (coal, underground and open-pit; ecol. impact of hard coal mining and river water pollution redn. at Petrosani coal basin, Romania)

IT Chemical oxygen demand

Economics

Suspended sediment
 (ecol. impact of hard coal mining and river water pollution redn. at Petrosani coal basin, Romania)

IT Carbonates, occurrence
 RL: POL (Pollutant); OCCU (Occurrence)
 (ecol. impact of hard coal mining and river water pollution redn. at Petrosani coal basin, Romania)

IT Coal, uses
 RL: NUU (Other use, unclassified); USES (Uses)
 (hard; ecol. impact of hard coal mining and river water pollution redn. at Petrosani coal basin, Romania)

IT Water pollution
 (river water; ecol. impact of hard coal mining and river water pollution redn. at Petrosani coal basin, Romania)

IT Chlorides, occurrence

Nitrites

Sulfates, occurrence
 RL: POL (Pollutant); OCCU (Occurrence)
 (river water; ecol. impact of hard coal mining and river water pollution redn. at Petrosani coal basin, Romania)

IT Solid wastes
 Solid wastes
 (tailings, coal; ecol. impact of hard coal mining and river water pollution redn. at Petrosani coal basin, Romania)

IT 12408-02-5, Hydrogen ion, occurrence
 RL: OCU (Occurrence, unclassified); OCCU (Occurrence)
 (ecol. impact of hard coal mining and river water pollution redn. at Petrosani coal basin, Romania)

IT 7439-89-6, Iron, occurrence
 RL: POL (Pollutant); OCCU (Occurrence)
 (river water total; ecol. impact of hard coal mining and river water pollution redn. at Petrosani coal basin, Romania)

IT 7782-44-7, Oxygen, occurrence
 RL: OCU (Occurrence, unclassified); OCCU (Occurrence)
 (river water; ecol. impact of hard coal mining and river water pollution redn. at Petrosani coal basin, Romania)

IT 7439-95-4, Magnesium, occurrence 7440-23-5, Sodium, occurrence
7440-43-9, Cadmium, occurrence 7440-70-2, Calcium, occurrence
14798-03-9, Ammonium, occurrence 14996-02-2, Hydrogen sulfate, occurrence
 RL: POL (Pollutant); OCCU (Occurrence)
 (river water; ecol. impact of hard coal mining and river water pollution redn. at Petrosani coal basin, Romania)

L2 ANSWER 9 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

Full Text Citing References

AN 1998:287095 CAPLUS
 DN 129:6453

h eb c g cg b cg

eb

ED Entered STN: 16 May 1998
 TI Manufacturing method of **smokeless** low-rank **coal** briquets
 IN Hongo, Takashi; Doi, Shigeyuki; Suetsugu, Kenji
 PA Ube Industries, Ltd., Japan
 SO Jpn. Kokai Tokkyo Koho, 6 pp.
 CODEN: JKXXAF

DT Patent
 LA Japanese
 IC ICM C10L005-06
 CC 51-24 (Fossil Fuels, Derivatives, and Related Products)
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	<u>JP 10121073</u>	A2	19980512	<u>JP 1996-279182</u>	19961022
PRAI	<u>JP 1996-279182</u>		19961022		

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
JP 10121073	ICM	C10L005-06

AB A method for manuf. of **smokeless** low-rank **coal** briquets comprises (1) formulating a mixt. of low-rank **coal** and S-fixation agent or clay minerals with water, and pulverizing-mixing to form a mixt. of main body having geometric mean size 100-500 μm , total water content 25-36%, and bulk d. 0.5-1.0 g/cm³ during forming, (2) formulating a mixt. of charcoal and S-fixation agent with binder and water, and pulverizing-mixing to form an igniting mixt. having geometric mean size 75-150 μm , total water content 28-36%, and bulk d. 0.2-0.7 g/cm³ during forming, and (3) compression molding the mixt. having the igniting mixt. on top of the main-body mixt.

ST **coal** briquet manuf low rank **smokeless**

IT Bentonite, uses

RL: NUU (Other use, unclassified); TEM (Technical or engineered material use); USES (Uses)

(binder; in manufg. method of **smokeless** low-rank **coal** briquets)

IT Clay minerals

RL: MOA (Modifier or additive use); NUU (Other use, unclassified); USES (Uses)

(in manufg. method of **smokeless** low-rank **coal** briquets)

IT Charcoal

RL: NUU (Other use, unclassified); TEM (Technical or engineered material use); USES (Uses)

(in manufg. method of **smokeless** low-rank **coal** briquets)

IT Combustion

Fuel briquets

(manufg. method of **smokeless** low-rank **coal** briquets)

IT 1305-62-0, Slaked lime, uses

RL: MOA (Modifier or additive use); NUU (Other use, unclassified); USES (Uses)

(S-fixation agent; manufg. method of **smokeless** low-rank **coal** briquets)

IT 9002-89-5, Polyvinyl alcohol

RL: NUU (Other use, unclassified); TEM (Technical or engineered material use); USES (Uses)

(binder; in manufg. method of **smokeless** low-rank **coal** briquets)

IT 497-19-8, Soda ash, uses 584-08-7, Potassium **carbonate**

RL: MOA (Modifier or additive use); NUU (Other use, unclassified); USES (Uses)

(in manufg. method of **smokeless** low-rank **coal**

briquets)

L2 ANSWER 10 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

Full Citing
 Text References

AN 1998:175999 CAPLUS
 DN 128:219340
 ED Entered STN: 25 Mar 1998
 TI Method of refining waste oils (petroleum products)
 IN Dimitrieva, Zinaida T.
 PA Destiny Oil Anstalt, Liechtenstein; Dimitrieva, Zinaida T.
 SO PCT Int. Appl., 28 pp.

CODEN: PIXXD2

DT Patent

LA English

IC ICM C10M175-00

CC 51-8 (Fossil Fuels, Derivatives, and Related Products)

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	<u>WO 9810045</u>	A1	19980312	<u>WO 1996-IB906</u>	19960909
	W: AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CU, CZ, DE, DK, EE, ES, FI, GB, GE, HU, IL, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, TJ, TM, TR, TT, UA, UG, US, UZ, VN, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM RW: KE, LS, MW, SD, SZ, UG, AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG				
	<u>AU 9667520</u>	A1	19980326	<u>AU 1996-67520</u>	19960909
PRAI	<u>WO 1996-IB906</u>		19960909		

CLASS

PATENT NO. CLASS PATENT FAMILY CLASSIFICATION CODES

WO 9810045	ICM	C10M175-00
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AB The invention relates to **ecol.** of the environment and conservation of energy-contg. mineral hydrocarbon raw material due to secondary use after treatment of the refined oils and cutting oils in the metal-working industry and in machine engineering. The method involves intensive treatment, without heating, of the raw material by extractant-absorbents as aq. solns. of phosphoric, hydrochloric and sulfuric acids, caustic soda at concns. from 2 up to 88% by mass with a volumetric ratio for the extractant:raw material from 1:1 up to 1:60, without drying or subsequent drying and neutralization over oxides, hydroxides, salts of alk. and alkali-earth metals. The method includes treatment of the raw material with the adsorbent where the raw material is filtered through charcoal or hard **coal** or coke cut with sand in a mass ratio from 1:0.5 to 1:45 and particle dispersion of 160-400 μm .

ST refining waste oil petroleum product

IT Petroleum refining

(extn.; waste oil refining by acid extn. and absorption process)

IT Wastes

(oil; waste oil refining by acid extn. and absorption process)

IT Lubricating oils

(used; waste oil refining by acid extn. and absorption process)

IT Absorbents

Coagulants

Drying

Drying agents

Extractants

Filtration

Neutralization

(waste oil refining by acid extn. and absorption process)

IT Alkali metal salts

h eb c g cg b cg

eb

Alkaline earth salts
 Bases, processes
 Charcoal
 Coke
 Hydroxides (inorganic)
 Oxides (inorganic), processes
 Zeolites (synthetic), processes
 RL: PEP (Physical, engineering or chemical process); PROC (Process)
 (waste oil refining by acid extrn. and absorption process)
 IT 497-19-8, Sodium **carbonate** (Na₂CO₃), processes 1305-78-8,
 Calcium oxide (CaO), processes 1310-58-3, Potassium hydroxide (KOH),
 processes 1310-73-2, Sodium hydroxide (NaOH), processes 7487-88-9,
 Sulfuric acid magnesium salt (1:1), processes 7601-54-9, Sodium
 phosphate (Na₃PO₄) 7647-01-0, Hydrochloric acid, processes 7664-38-2,
 Phosphoric acid, processes 7664-93-9, Sulfuric acid, processes
7732-18-5, Water, processes 7757-82-6, Sodium sulfate (Na₂SO₄),
 processes 10043-52-4, Calcium chloride (CaCl₂), processes
 RL: PEP (Physical, engineering or chemical process); PROC (Process)
 (waste oil refining by acid extrn. and absorption process)

RE.CNT 11 THERE ARE 11 CITED REFERENCES AVAILABLE FOR THIS RECORD

RE

- (1) Agnew, J; US 4491515 A 1985 CAPLUS
- (2) Carlos, D; US 3376216 A 1968 CAPLUS
- (3) Coal Chem Inst Res; SU 1599420 A 1990 CAPLUS
- (4) Deutsche Gasolin Aktiengesellschaft; GB 445731 A 1936 CAPLUS
- (5) Deutsche Gasolin Aktiengesellschaft; GB 451760 A 1936 CAPLUS
- (6) Fainman, M; US 3819508 A 1974 CAPLUS
- (7) Fainman, M; US 3835035 A 1974 CAPLUS
- (8) Godfrey L Cabot Inc; GB 799278 A 1958
- (9) Veb Hydrierwerk Zeitz; DD 267258 A 1989 CAPLUS
- (10) Veb Pck Schwedt; DD 212528 A 1984 CAPLUS
- (11) Willis; GB 189701 A 1922 CAPLUS

L2 ANSWER 11 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

Full	Citing
Text	References

AN 1997:157180 CAPLUS
 DN 126:201472
 ED Entered STN: 10 Mar 1997
 TI Upgrading selected Czech coals for home and industrial heating
 AU Musich, Mark A.; Young, Brian C.
 CS Energy & Environmental Research Center, University of North Dakota, Grand
 Forks, ND, 58202, USA
 SO Proceedings - Institute for Briquetting and Agglomeration, Biennial
 Conference (1996), Volume Date 1995, 24, 103-116
 CODEN: PIBABP; ISSN: 0145-8701
 PB Institute for Briquetting and Agglomeration
 DT Journal
 LA English
 CC 51-17 (Fossil Fuels, Derivatives, and Related Products)
 Section cross-reference(s): 59
 AB The Czech Republic has large **coal** reserves, particularly brown **coal**
 and lignite, and to a lesser extent, bituminous **coal**. Concurrent with
 the recent political changes, there has been a reassessment of the role of
coal for elec. and heating energy in the future economy, owing to the
 large dependence on brown **coal** and lignite and the implementation of
 more stringent environmental regulations. These **coals** have a relatively
 high sulfur content, typically 1-3 wt%, and ash content, leading to
 significant SO₂ and other gaseous and particulate emissions. Some of the
 bituminous **coals** also exhibit high ash content. Against this
 background, the Energy & Environmental Research Center, on behalf of the
 U.S. Agency for International Development and the U.S. Department of
 Energy Office of Fossil Energy, undertook a project on upgrading Czech
coals to achieve desired fuel properties. The purpose of the project

was to assist the city of Usti and Labem in Northern Bohemia in developing cost-effective alternatives for reducing environmental emissions from district and home heating systems. Three **coals** were selected, namely Bilina and Nastup lignites from Northern Bohemia and Ostrava bituminous **coal** from Moravia, for a limited tech. investigation to assess their potential for upgrading. All **coals** were analyzed for ash and sulfur content, forms of sulfur, and ash compn. Bilina and Nastup lignites were subjected to wet and dry phys. cleaning methods to reduce the ash and sulfur content. Phys. cleaned Bilina lignite and raw Ostrava bituminous **coal** were carbonized to reduce volatile matter content. Selected phys. cleaned and carbonized **coals** were tableted with a starch binder and calcium **carbonate**, the latter being added for sulfur capture. Following anal. of the tableted **coal** products, the latter were evaluated for their potential application as district and home heating fuels. Fuels prepnd. from raw Ostrava bituminous **coal** and Bilina lignite cleaned by the float-sink method would be acceptable for steam-raising in com. and energy installation applications. Tableted fuels prepnd. from Ostrava bituminous **coal** and a carbonized, magnetically cleaned Bilina product would be suitable as **smokeless** fuels for home heating because of their reduced sulfur and volatile matter contents. The Bilina lignite-derived fuels contg. the sulfur capture agent have an emission-effective sulfur content similar to that of the Ostrava fuels. The results of this project indicate the opportunities for reducing sulfur dioxide, smoke, and particulate emissions in Northern Bohemia, esp. around Usti and Labem, by the proper cleaning treatment, briquetting, and utilization of Czech indigenous **coals**. However, further development at pilot and demonstration scales is required, along with an economic evaluation of the fuel prepnd. and fuel combustion systems for heating.

- ST coal lignite upgrading home industrial heating; treatment coal lignite home industrial heating; beneficiation coal lignite steam raising; deashing briquetting lignite; sulfur dioxide emission control lignite prepnd; smoke particulate emission control lignite cleaning
- IT Coal treatment
 - (cleaning; upgrading selected Czech coals for home and industrial heating)
- IT Coal treatment
 - (deashing; upgrading selected Czech coals for home and industrial heating)
- IT Airborne particles
 - Smoke
 - (emission control; upgrading selected Czech coals for home and industrial heating)
- IT Carbonization
 - (of lignite; upgrading selected Czech coals for home and industrial heating)
- IT Heating
 - (steam; upgrading selected Czech coals for home and industrial heating)
- IT Heating
 - (upgrading selected Czech coals for)
- IT Coal treatment
 - Fuel briquets
 - Tablets
 - (upgrading selected Czech coals for home and industrial heating)
- IT Coal, uses
 - Lignite
 - RL: PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)
 - (upgrading selected Czech coals for home and industrial heating)
- IT 9005-25-8, Starch, uses
 - RL: NUU (Other use, unclassified); USES (Uses)
 - (in coal briquetting; upgrading selected Czech coals for home and industrial heating)
- IT 471-34-1, Calcium **carbonate**, uses
 - RL: NUU (Other use, unclassified); USES (Uses)

(in coal prepn., for sulfur capture; upgrading selected Czech coals for home and industrial heating)

RE.CNT 8 THERE ARE 8 CITED REFERENCES AVAILABLE FOR THIS RECORD

RE

- (1) Couch, G; Lignite Upgrading, 1990, P73
- (2) Epa; 1992, EPA/430/R-92/1008
- (3) Iea; IEA report 1994, P165
- (4) Krivsky, Z; Private Communication 1994
- (5) Mall, L; Proceedings of the Production and the Utilization of Ecological Fuels from East Central Coals Workshop 1994, P13
- (6) Ministry Of Industry And Trade; Energy Policies in the Czech Republic, Section 2.5.1, Coal In-Country Fuel Resources Quality and Ecological Characteristics, 1994
- (7) Moldan, B; Environ Sci & Technol 1992, V26(1), P14 CAPLUS
- (8) Young, B; 1995, 95-EERC-05-01, P47

L2 ANSWER 12 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

<input type="checkbox"/> Full	<input checked="" type="checkbox"/> Citing
<input type="checkbox"/> Text	<input type="checkbox"/> References

AN 1996:598368 CAPLUS
 DN 125:252497
 ED Entered STN: 07 Oct 1996
 TI Major chemical components and mineralogical character of the inorganic matter of energetic coals from the east part of Upper Silesia Coal Basin (Poland)
 AU Bojarska, Katarzyna; Bzowski, Zbigniew
 CS Pol.
 SO Proceedings - Annual International Pittsburgh Coal Conference (1995), 12th, 511-516
 CODEN: PICNE4; ISSN: 1075-7961
 PB Pittsburgh Coal Conference, University of Pittsburgh
 DT Journal
 LA English
 CC 51-16 (Fossil Fuels, Derivatives, and Related Products)
 Section cross-reference(s): 59, 60, 61
 AB It was confirmed that the major components of the mineral matter of the **coals** are clay minerals and **carbonates**. The results of mineralogical and chem. studies of the energetic **coals** are used in **ecol.** evaluation of ash utilization.
 ST coal mineral matter ash utilization environment
 IT Environmental pollution
 Water pollution
 (by coal ash, control of; mineral matter of energetic coals of Upper Silesian Coal Basin (Poland))
 IT Environment
 (mineral matter of energetic coals of Upper Silesian Coal Basin (Poland))
 IT Clay minerals
 Minerals
 RL: OCU (Occurrence, unclassified); POL (Pollutant); OCCU (Occurrence)
 (mineral matter of energetic coals of Upper Silesian Coal Basin (Poland))
 IT Ashes (residues)
 (coal, environment and utilization of; mineral matter of energetic coals of Upper Silesian Coal Basin (Poland))

L2 ANSWER 13 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

<input type="checkbox"/> Full	<input checked="" type="checkbox"/> Citing
<input type="checkbox"/> Text	<input type="checkbox"/> References

AN 1996:43533 CAPLUS
 DN 124:122448
 ED Entered STN: 23 Jan 1996
 TI Late Miocene environmental change in Nepal and the northern Indian subcontinent: stable isotopic evidence from paleosols

AU Quade, Jay; Cater, John M. L.; Ojha, Tank P.; Adam, Jon; Harrison, T. Mark
 CS Department of Geosciences, University of Arizona, Tucson, AZ, 85721, USA
 SO Geological Society of America Bulletin (1995), 107(12), 1381-97
 CODEN: BUGMAF; ISSN: 0016-7606
 PB Geological Society of America
 DT Journal
 LA English
 CC 53-6 (Mineralogical and Geological Chemistry)
 AB Neogene sediments belonging to the Siwalik Group crop out in the Himalayan foothills along the length of southern Nepal. Carbon and oxygen isotopic analyses of Siwalik paleosols from four long Siwalik sections record major **ecol.** changes over the past ~11 Myr. The carbon isotopic compn. of both soil **carbonate** and org. matter shifts dramatically starting ~ 7.0 Myr, marking the displacement of largely C3 vegetation, probably semi-deciduous forest, by C4 grasslands. By the beginning of the Pliocene, all the flood plains of major rivers in this region were dominated by monsoonal grasslands. The floral shift away from woody plants is also reflected by the decline and final disappearance of fossil leaves and the decrease in **coal** logs in the latest Miocene. A similar carbon isotopic shift has been documented in the paleosol and fossil tooth record of Pakistan, and in terrigenous org. matter from the Bengal Fan, showing that the floral shift was probably continent-wide. The latest Miocene also witnessed an av. change of ~4.permill. in the oxygen isotopic compn. of soil **carbonate**, as obsd. previously in Pakistan. The reasons for this are unclear; if not diagenetic, a major environmental change is indicated, perhaps related to that driving the carbon isotopic shift. Recently described pollen and leaf fossils from the Surai Khola section show that evergreen forest was gradually displaced by semi-deciduous and dry deciduous forest between 11 and 6 Myr. The gradual nature of this floral shift, which culminated in the rapid expansion of C4 grasses starting ~7.0 Myr ago, is difficult to explain by a decrease in atm. pCO₂ alone (Cerling et al., 1993) but fits well with a gradual onset of monsoonal conditions in the late Miocene in the northern Indian subcontinent. Himalayan uplift, driving both monsoonal intensification and consumption of CO₂ through weathering, may be the common cause behind major late Miocene environmental change globally. However, the decline of effective moisture assocd. with monsoon development has probably slowed, not increased, the rate of consumption of CO₂ by chem. weathering of Himalayan sediments.
 ST paleosol carbon isotope Miocene paleoenvironment Nepal; oxygen isotope **carbonate** paleosol paleoclimate Nepal
 IT Plant
 (carbon- and oxygen-isotope geochem. of paleosols as evidence of late Miocene environmental change in Nepal and northern Indian subcontinent from semi-deciduous forest to grasslands)
 IT Climate
 Environment
 (paleo-, carbon- and oxygen-isotope geochem. of paleosols as evidence of late Miocene environmental change in Nepal and the northern Indian subcontinent)
 IT Soils
 (paleosols, carbon- and oxygen-isotope geochem. of paleosols as evidence of late Miocene environmental change in Nepal and the northern Indian subcontinent)
 IT 14762-74-4, Carbon-13, occurrence
 RL: GOC (Geological or astronomical occurrence); GPR (Geological or astronomical process); OCCU (Occurrence); PROC (Process)
 (geol. indicator; in **carbonate** and org. matter from paleosols as evidence of late Miocene environmental change in Nepal and the northern Indian subcontinent)
 IT 14797-71-8, Oxygen-18, occurrence
 RL: GOC (Geological or astronomical occurrence); GPR (Geological or astronomical process); OCCU (Occurrence); PROC (Process)
 (geol. indicator; in **carbonate** from paleosols as evidence of

late Miocene environmental change in Nepal and the northern Indian subcontinent)

L2 ANSWER 14 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

Full Text Citing References

AN 1992:534600 CAPLUS
 DN 117:134600
 ED Entered STN: 04 Oct 1992
 TI Is Gaia endothermic?
 AU Hsue, K. J.
 CS Geol. Inst., ETH-Zentrum, Zurich, CH-8092, Switz.
 SO Geological Magazine (1992), 129(2), 129-41
 CODEN: GEMGA4; ISSN: 0016-7568
 DT Journal; General Review
 LA English
 CC 53-0 (Mineralogical and Geological Chemistry)
 AB A review with 42 refs. discusses the uniqueness of the earth, its atm., biol., and climatol. evolution, as well as the interrelatedness between them. Geol. evidence suggests that Gaia is endothermic: her body temp. has varied, but within limits; there has been no runaway greenhouse like Venus, nor deep freeze like Mars. This paper presents a hypothesis that the Earth's climate has been ameliorated by living organisms: they have served either as heaters or air-conditioners, and their **ecol.** tolerance is the sensor of Gaia's thermostat. At the beginning, 3.8 or 3.5 Gyr ago, only anaerobic autotrophs capable of tolerating high temps. thinned out the atm. CO₂ through carbon fixation. Fossil org. carbon was utilized by anaerobic heterotrophs to reinforce the effectiveness of the Late Archean greenhouse, when solar luminosity was weaker than it is now. With the increasing solar luminosity during early Proterozoic time, new life forms such as cyanobacteria evolved, removing CO₂ from the atm. and storing it in stromatolitic **carbonates**. Over-eager cyanobacteria may have consumed too much greenhouse CO₂ to cause glaciation. Their decline coincided in timing with the rise of the Ediacaran faunas which had no **carbonate** skeletons. The change in the mode of carbon-cycling may have started the warming trend after the Proterozoic glaciation. The Cambrian explosion was an event when skeletal eukaryotes usurped the function of prokaryotes in removing greenhouse CO₂ through CaCO₃ pptn. With the evolution of land plants, **coal**-makers took over the 'air-conditioning' duty. They over-did it, and Permo-Carboniferous glaciation ensued. After a wholesale turnover of the faunas and floras at the end of the Paleozoic, more CO₂ was released than fixed in Early Mesozoic time. The warming trend reached its zenith in the early Cretaceous, when flowering trees and calcareous plankton began to flourish. The decline since then, with a temporary restoration during Early Paleogene time, could be a manifestation of the varying efficiency of extg. and burying carbon dioxide, in the form of inorg. and org. carbon. The relation of atm. CO₂ and climatic variation is documented by study of air bubbles in ice cores. Yet there is also correlation to astronomical cycles. The latter seem to have triggered changes which are amplified by feedback mechanisms of carbon cycling.
 ST review carbon cycle climate Earth history; atm biol climatol evolution
 Earth review
 IT Evolution
 (effect on geochem. carbon cycle, paleoclimate in relation to)
 IT Atmosphere
 (evolution of, biol.- and climatol. evolution and geochem. cycle in
 relation to)
 IT Climate
 (paleo-, atm.- and biol. evolution and the carbon geochem. cycle in
 relation to)
 IT 7440-44-0, Carbon, properties
 RL: PRP (Properties)
 (geochem. cycle, biol. and paleoclimatol. evolution in relation to,
 during Earth's history)

L2 ANSWER 15 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

Full	Citing
Text	References

AN 1985:223291 CAPLUS
 DN 102:223291
 ED Entered STN: 29 Jun 1985
 TI Coal or coke briquets
 IN Messenig, Leo; Cieslik, Wolfgang; Opdenwinkel, Heinz
 PA Ruhrkohle A.-G., Fed. Rep. Ger.
 SO Eur. Pat. Appl., 13 pp.
 CODEN: EPXXDW
 DT Patent
 LA German
 IC ICM C10L005-14
 CC 51-24 (Fossil Fuels, Derivatives, and Related Products)

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
<u>PI</u>	<u>EP 135785</u>	A2	19850403	<u>EP 1984-109823</u>	19840817
	<u>EP 135785</u>	A3	19870114		
	<u>EP 135785</u>	B1	19891025		
	R: AT, BE, CH, DE, FR, GB, IT, LI, LU, NL, SE				
	<u>DE 3335240</u>	A1	19850418	<u>DE 1983-3335240</u>	19830929
	<u>AT 47602</u>	E	19891115	<u>AT 1984-109823</u>	19840817
<u>PRAI</u>	<u>DE 1983-3335240</u>		19830929		
	<u>EP 1984-109823</u>		19840817		

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
<u>EP 135785</u>	ICM	C10L005-14

AB **Smokeless** sootless mech. strong **coal** or coke briquets are manufd. by spraying a finely divided fuel (at 80-95°) with a hot soln. of fully saponified (>97%) poly(vinyl acetate) (viscosity 40-100 mPa-S) binding agent, homogenizing the mixt., and pressing the fuel into briquets with dewatering. The fuel briquets contain 0.3-2 wt.% poly(vinyl alc.). In addn., <3 wt.% molasses or <1.5 wt.% CaCO₃ can be added to improve the heat stability.
 ST coal coke briquet molasses; polyvinyl alc coal coke briquet; binding material coal coke briquet; calcium **carbonate** coal coke briquet
 IT Fuel briquets
 (coal and coke, manuf. of, saponified poly(vinyl acetate) binders for, molasses or calcium **carbonate** additives for)
 IT Molasses
 (**smokeless** sootless mech. strong **coal** or coke briquets contg.)
 IT 471-34-1, uses and miscellaneous 9003-20-7D, saponified.
 RL: USES (Uses)
 (**smokeless** sootless mech. strong **coal** or coke briquets contg.)

L2 ANSWER 16 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

Full	Citing
Text	References

AN 1985:206483 CAPLUS
 DN 102:206483
 ED Entered STN: 15 Jun 1985
 TI Coal and coke briquets
 IN Messenig, Leo; Cieslik, Wolfgang; Opdenwinkel, Heinz
 PA Ruhrkohle A.-G., Fed. Rep. Ger.
 SO Ger. Offen., 14 pp.
 CODEN: GWXXBX
 DT Patent
 LA German

IC ICM C10L005-10
 CC 51-24 (Fossil Fuels, Derivatives, and Related Products)

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	<u>DE 3335241</u>	A1	19850418	<u>DE 1983-3335241</u>	19830929
PRAI	<u>DE 1983-3335241</u>		19830929		

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
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DE 3335241	ICM	C10L005-10
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AB **Smokeless** spotless mech. strong **coal** or coke briquets contain 1.2-1.8 wt.% CaCO₃ and 0.5-1.5 wt.% fully sapond. poly(vinyl acetate) with viscosity 50-70 mPa-s. The finely ground **coal** or coke is heated to 100 ± 20° and then sprayed with hot (heated to 80-95°) aq. soln. of poly(vinyl alc.) contg. CaCO₃. The mixt. is then homogenized and pressed and dewatered to briquets.

ST **coal** coke briquetting **smokeless** spotless; calcium **carbonate coal** coke briquet; polyvinyl alc **coal** coke briquet

IT Fuel briquets
 (coal or coke, manuf. of, **smokeless** sootless,
 additives for)

IT 471-34-1, uses and miscellaneous 9003-20-7D, sapond.

RL: USES (Uses)
 (additives, in manuf. of **smokeless** sootless mech. strong
coal or coke briquets)

L2 ANSWER 17 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

Full Text	Citing References
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AN 1974:493906 CAPLUS

DN 81:93906

ED Entered STN: 12 May 1984

TI Briquet for frost-prevention

IN Onozawa, Tatsugoro

SO Jpn. Tokkyo Koho, 2 pp.

CODEN: JAXXAD

DT Patent

LA Japanese

IC C10L

CC 51-23 (Fossil Fuels, Derivatives, and Related Products)

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI <u>JP 49011443</u>	B4	19740316	<u>JP 1970-74594</u>	19700827
PRAI <u>JP 1970-74594</u>		19700827		

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
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JP 49011443	IC	C10L
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AB A briquet is prep'd. contg. carbonaceous material and (NH₄)₂CO₃ or NH₄HCO₃. It has no harmful effect on plant life. The carbonaceous material, e.g. **smokeless coal**, coke, or charcoal, is mixed with (NH₄)₂CO₃ or NH₄HCO₃ 5-20, CaCO₃ 3-5, iron powder 1, and binder ~2%. Water is mixed in and the briquets are formed and dried. Thus, carbonaceous material contg. 7 parts **smokeless coal** per 3 parts charcoal 82, (NH₄)₂CO₃ 10, CaCO₃ 5, iron powder 1, and dextrin 2 wt. parts were mixed with water and briquetted. A 1450-g briquet emitted a white smoke contg. NH₃ for 3 hr during combustion. The briquet continued to burn for an addnl. 4 hr without generating smoke.

ST frost damage preventive briquet; plant frost damage prevention

IT Briquets, fuel
 (charcoal-coal, contg. ammonium **carbonate**)

IT Frost

h ebc g cg b cg

eb

(crop damage from, fuel briquets for prevention of)

- IT Charcoal
 RL: USES (Uses)
 (fuel briquets contg.)
 IT 471-34-1, uses and miscellaneous 506-87-6 9004-53-9
 RL: USES (Uses)
 (fuel briquets contg.)
 IT 7439-89-6, uses and miscellaneous
 RL: USES (Uses)
 (powd., fuel briquets contg.)

L2 ANSWER 18 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

 Citing
 References

AN 1963:453746 CAPLUS
 DN 59:53746
 OREF 59:9697c-e
 ED Entered STN: 22 Apr 2001
 TI Fuel briquets
 PA Great Lakes Carbon Corp.

SO 5 pp.

DT Patent

LA Unavailable

CC 26 (Coal and Coal Derivatives)

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI GB 929810		19630626	GB	19610327

CLASS

PATENT NO.	CLASS	PATENT FAMILY CLASSIFICATION CODES
GB 929810		

AB Bituminous coal particles with 14-23% volatile content are heated at a rate in excess of 2000°F. surface temp. per sec. to about 1600°F. in a gas stream contg. 2-4 cu. ft. of O per lb. of coal. The sepd. solid has a particle size between 5 and 1000 μ and a volatile content of 7-15%. If desirable, the treatment may be repeated further to depress the volatile content. The powder is mixed with an aq. binder contg. a starch, briquetted, and dried. Various substances may be incorporated in the mixt. before pressing to enhance certain properties. These include alkali carbonates, nitrates, chlorates, chlorides, and the oxides of Cu or Pb, or lime and hickory sawdust. The briquets are smokeless. Coating them with a starch soln. reduces the amt. of fines on the surface. Cf. U.S. 2,017,402 (CA 29, 82949); Brit. 786,207.

IT Briquets, fuel
 (from flash-calcined coal)

L2 ANSWER 19 OF 19 CAPLUS COPYRIGHT 2004 ACS on STN

 Citing
 References

AN 1935:4627 CAPLUS
 DN 29:4627
 OREF 29:575h-i,576e-i,577a-b
 ED Entered STN: 16 Dec 2001
 TI The solid products of the carbonization of coal
 CS Chemical Department, South Metropolitan Gas Co.

SO South Metropolitan Gas Co., London (1934) 123 pp.

DT Journal

LA Unavailable

CC 21 (Fuels, Gas, Tar, and Coke)

AB A comparison of the results of lab. tests of ignition temp., combustibility in air, and reactivity with CO₂ and steam on a series of cokes and chars showed good general correlation among the 4 different tests and indicated that any one of them would serve to evaluate the burning properties of cokes in grate fires. Several coals were

carbonized in a 6-in. Carborundum retort at various temps. ranging from 500° to 1050°. The ignition temps. of the resulting cokes increased from 470° to 626°, as the carbonizing temp. increased from 650° to 900°. Up to 900°, the H content of the cokes decreased in almost direct proportion to the increase of carbonizing temp. In general, the effect of increasing carbonizing temp. on the properties of the cokes was to decrease the reactivity to air, CO₂, steam, and H₂SO₄, and to increase the apparent and true density, and the elec. cond. These changes in coke characteristics were most marked at about 700° carbonization temp. and coincided with the change in appearance from a dull black to the characteristic silvery sheen of high-temp. coke. The absorptive capacity of the cokes for CO₂ increased sharply with increasing temp. of carbonization up to a max. at 700° and then fell rapidly. Deposition of graphitic C reduced slightly the reactivity of the high-temp. cokes but seemingly was not the major cause of the large difference between low- and high-temp. cokes. Alteration of the phys. structure produced by any modification of the high-temp. carbonizing process likely to be practicable was quite ineffective in producing a substantial improvement in the combustibility of the coke. Addn. of Na **carbonate** increased the reactivity with steam but did not reduce the ignition temp. proportionately nor did it measurably increase the combustibility in air. As a result of this investigation the authors believe that the difference in reactivity between low- and high-temp. coke is due to a profound change in chem. rather than phys. structure. Assuming that the benzene ring nucleus plays an important part in the chem. structure of **coal**, they believe that in the low-temp. carbonization of **coal**, i. e., below 700° side chains are stripped from the nucleus without coalescence of these nuclei; whereas, in carbonization above 700°, the single nuclei coalesce to form complex polycyclic compds. coincident with a marked evolution of H. The closer assocn. of the C atoms without the frequent interposition of H atoms suggests a plausible explanation of the reduced chem. activity. To obtain a **smokeless** solid fuel, readily ignitable and freely burning in the ordinary domestic grate the **coal** must be carbonized below 700°. This fuel should have a low ash, not over 5% moisture, about 13% volatile matter, elec. resistivity not less than one million ohms, and be between 1 and 3 in. in size. An exptl. study of the relation of the characteristics of high-temp. cokes in relation to its use in gas grates and hot-water boilers showed that fairly satisfactory results could be had by the use of specially designed "Metro" coke grates and by preparing the coke as follows: size, 1-2 in., moisture 3%, volatile therms per ton 2 1/2, and as low ash vol. as possible. For hot-water boilers, the min. rate of combustion at which the fire will maintain itself is the most important property.

IT Char
 (burning properties of, carbonizing conditions and)
 IT Coke
 (combustibility of, effect of rate of heating of coal on)
 IT Fuels
 (smokeless, by low-temp. carbonization)
 IT Carbonization
 (solid products of)

```
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                                                ENTRY        SESSION
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=> s ecological coal
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    4 ECOLOGICALS
  9918 ECOLOGICAL
    (ECOLOGICAL OR ECOLOGICALS)
  34118 ECOL
    1 ECOLS
  34119 ECOL
    (ECOL OR ECOLS)
  38406 ECOLOGICAL
    (ECOLOGICAL OR ECOL)
  209512 COAL
  35522 COALS
  211340 COAL
    (COAL OR COALS)
L3      10 ECOLOGICAL COAL
    (ECOLOGICAL(W) COAL)
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=> d 13 1-10 all

L3 ANSWER 1 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN

Full	Citing
Text	References

AN 2003:504083 CAPLUS
 DN 139:194292
 ED Entered STN: 02 Jul 2003
 TI The microbial ecology of soil surrounding an outdoor coal storage pile
 AU Dore, Sophia Yasmine
 CS Univ. of Notre Dame, Notre Dame, IN, USA
 SO (2002) 128 pp. Avail.: UMI, Order No. DA3068011
 From: Diss. Abstr. Int., B 2003, 63(10), 4482
 DT Dissertation
 LA English
 CC 10-6 (Microbial, Algal, and Fungal Biochemistry)
 AB Unavailable
 ST microbial **ecol** coal storage pile soil
 IT Microbial ecology
 Soil pollution
 (microbial ecol. of soil surrounding outdoor coal storage pile)
 IT Coal, biological studies
 RL: BSU (Biological study, unclassified); BIOL (Biological study)
 (microbial ecol. of soil surrounding outdoor coal storage pile)
 IT Ecology
 (soil; microbial ecol. of soil surrounding outdoor coal storage pile)

L3 ANSWER 2 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN

Full	Citing
Text	References

AN 2001:740293 CAPLUS
 DN 136:234463
 ED Entered STN: 11 Oct 2001
 TI Clean and **ecological** coal combustion in the binary circulating fluidized bed
 AU Nowak, W.; Muskala, W.
 CS Technical University of Czestochowa, Czestocowa, 42-200, Pol.
 SO Energy (Oxford, United Kingdom) (2001), 26(12), 1109-1120
 CODEN: ENEYDS; ISSN: 0360-5442
 PB Elsevier Science Ltd.
 DT Journal
 LA English
 CC 51-18 (Fossil Fuels, Derivatives, and Related Products)
 AB The aim of this paper is to increase the understanding of the role of the binary circulating fluidized bed in the process of clean and **ecol.** coal combustion. The operating range of a stable fluidized bed, as a function of gas velocity changes and the flow rate of fine particles, is detd. for all possible conditions. Expts. concerning the combustion and desulfurization processes in multi-solid fluidized bed (MSFB) and circulating fluidized bed (CFB) systems give evidence that the residence time of burnt particles in the combustion chamber of MSFB is much extended. This is directly reflected in better combustion conditions, esp. those for fine particles, as well as in the process of desulfurization. The advantages of the binary circulating fluidized bed over typical circulating systems make it one of the most efficient methods of clean and **ecol.** coal combustion.
 ST coal combustion binary circulating fluidized bed; desulfurization coal
 binary circulating fluidized bed
 IT Combustion
 (clean and **ecol.** coal combustion in binary
 circulating fluidized bed)
 IT Coal, processes
 RL: CPS (Chemical process); PEP (Physical, engineering or chemical

process); PROC (Process)
 (clean and **ecol. coal** combustion in binary
 circulating fluidized bed)
 IT Coal treatment
 (desulfurization; in binary circulating fluidized bed)
 IT Fluidized beds
 (recirculating; clean and **ecol. coal** combustion in
 binary circulating fluidized bed)
 RE.CNT 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD
 RE
 (1) Kojima, Y; Circulating fluidized beds technology 1988, P369
 (2) Nowak, W; Gospodarka Paliwami i Energia 1995, V7, P2
 (3) Nowak, W; Proceedings of the 5th SCEJ Symposium on Circulating Fluidized
 Beds 1993, P14
 (4) Nowak, W; Proceedings of the International Conference on Energy Systems and
 Ecology 1993, P37
 (5) Win, K; J Chem Eng Jpn 1995, V28(5), P535 CAPLUS

L3 ANSWER 3 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN

	Full Text	Citing References
AN	2001:1584	CAPLUS
DN	134:58770	
ED	Entered STN:	01 Jan 2001
TI	The international coal market - yesterday, today, tomorrow	
AU	Stadelhofer, Jurgen W.	
CS	Vorstandes RAG Coal International AG, Essen, D-45128, Germany	
SO	Erzmetall (2000), 53(10), 613-623	
	CODEN: ERZMAK; ISSN: 0044-2658	
PB	GDMB-Informationsgesellschaft	
DT	Journal; General Review	
LA	German	
CC	51-0 (Fossil Fuels, Derivatives, and Related Products)	
AB	A discussion and review without refs. Coal currently covers about 25 % of primary energy consumption in the world, a share that will remain about const. over the coming 20 yr. In view of the expected increase of energy consumption, this means a significant rise of coal demand - by about 30 % in 2020. The principal fields of application are the power and the steel industries. The share of coal in the prodn. of elec. power amts. in certain countries to more than 70 % - in the USA 56 %. Coal reserves in the world are adequate to cope with the expected increase in consumption. The importance of the international coal trade is growing. Both the steel and the power industries have potentials for reducing their CO ₂ emissions. Following the purchase of companies in both the USA and Australia, RAG has become one of the leading international coal companies, and is actively engaged in developing clean coal technologies, with a view to ensuring ecol. coal utilization in future.	
ST	review coal economics	
IT	Economics (international coal market yesterday, today, and tomorrow)	
IT	Coal, uses (RL: TEM (Technical or engineered material use); USES (Uses) (international coal market yesterday, today, and tomorrow))	

L3 ANSWER 4 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN

	Full Text	Citing References
AN	1995:786407	CAPLUS
DN	123:265274	
ED	Entered STN:	12 Sep 1995
TI	Ecological and geochemical evaluation of tailings of the coal mining industry	
AU	Bulavina, A. V.; Bydtaeva, N. G.; Gavrilei, V. A.	
CS	Russia	

SO Razvedka i Okhrana Nedr (1995), (3), 37-8
 CODEN: RZONAV; ISSN: 0034-026X
 PB Nedra
 DT Journal
 LA Russian
 CC 60-6 (Waste Treatment and Disposal)
 Section cross-reference(s): 51, 53
 AB Anal. of tailings from coal mines of the Donets coal basin (Ukraine) indicated that the waste rock piles are the source of environmental pollution (of soil, water and biota) by highly toxic elements and compds. Exptl. results indicated that aq. solns. remove Se, Cu, V, Cr, and Zn from tailings, making these wastes unsuitable for use as fertilizers. An increased migration of some toxic elements, esp. such as S, Se, and F can cause pollution of drinking water sources. The assessment of the environmental effects of tailings storage would require a complex study of soil, water, and a biogenic component, with the identification of migration pathways and accumulation areas of toxic elements.
 ST ecol effect coal mining tailing; geochem ecol coal mining waste
 IT Environmental pollution
 Environmental transport
 Soil pollution
 Water pollution
 (ecol. and geochem. evaluation of tailings from coal mining)
 IT Trace elements, occurrence
 RL: POL (Pollutant); OCCU (Occurrence)
 (ecol. and geochem. evaluation of tailings from coal mining)
 IT Mines and Mining
 (coal, ecol. and geochem. evaluation of tailings from coal mining)
 IT Waste solids
 (tailings, ecol. and geochem. evaluation of tailings from coal mining)
 IT 7439-89-6, Iron, occurrence 7439-92-1, Lead, occurrence 7439-93-2,
 Lithium, occurrence 7439-97-6, Mercury, occurrence 7439-98-7,
 Molybdenum, occurrence 7440-02-0, Nickel, occurrence 7440-24-6,
 Strontium, occurrence 7440-31-5, Tin, occurrence 7440-38-2, Arsenic,
 occurrence 7440-41-7, Beryllium, occurrence 7440-42-8, Boron,
 occurrence 7440-47-3, Chromium, occurrence 7440-48-4, Cobalt,
 occurrence 7440-50-8, Copper, occurrence 7440-62-2, Vanadium,
 occurrence 7440-66-6, Zinc, occurrence 7440-69-9, Bismuth, occurrence
 7704-34-9, Sulfur, occurrence 7782-49-2, Selenium, occurrence
 16984-48-8, Fluoride, occurrence
 RL: POL (Pollutant); OCCU (Occurrence)
 (ecol. and geochem. evaluation of tailings from coal mining)

L3 ANSWER 5 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN

	Full Text	Citing References
AN	1995:745081	CAPLUS
DN	123:148605	
ED	Entered STN:	18 Aug 1995
TI	Ecological and economical aspects of coal combustion as a function of its quality parameters	
AU	Kurczabinski, Leon	
CS	Pol.	
SO	Przeglad Gorniczy (1995), 51(4), 46-52	
	CODEN: PRGOAI; ISSN: 0033-216X	
PB	Wydawnictwo SIGMA-NOT	
DT	Journal	
LA	Polish	
CC	51-18 (Fossil Fuels, Derivatives, and Related Products)	
	Section cross-reference(s): 59	
AB	The state and directions of development of ecol. coal utilization are described taking into consideration both Polish and foreign research and investment programs. The problems presented by the emission of pollutants to the atm. by the Polish power industry are discussed. An account is	

given of the effect of application of high quality coals yielded by mech. coal prepns. methods, on the economics of certain operations assocd. with power prodn. and with environmental protection.

ST coal combustion power plant emission Poland

IT Air pollution
(ecol. and economical aspects of coal combustion and emission in relation to Polish power industry)

IT Combustion
(of coal; ecol. and economical aspects of coal combustion as a function of its quality parameters)

IT Power
(plant; ecol. and economical aspects of coal combustion and emission in relation to Polish power industry)

L3 ANSWER 6 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN

Full Text Citing References

AN 1986:36448 CAPLUS
 DN 104:36448
 ED Entered STN: 08 Feb 1986
 TI Coal-water fuel combustion; fundamentals and application. A North American overview
 AU Beer, J. M.
 CS Massachusetts Inst. Technol., Cambridge, MA, USA
 SO Institution of Chemical Engineers Symposium Series (1985), 95(Eur. Conf. Coal Liq. Mixtures, 2nd), 377-405
 CODEN: ICESDB; ISSN: 0307-0492
 DT Journal; General Review
 LA English
 CC 51-0 (Fossil Fuels, Derivatives, and Related Products)
 AB A review with 51 refs. The properties of coal-water slurries (CWS), burner development for CWS combustion, ecol. advantages of CWS over powd. coal combustion, and economic considerations are included.
 ST review coal water slurry combustion; burner coal water slurry review; ecol coal water slurry review; economics coal water slurry review
 IT Environmental pollution
 (furnace firing with coal-water slurries effect on)
 IT Combustion
 (of coal-water slurries)
 IT Firing of furnaces
 (with coal-water slurries)

L3 ANSWER 7 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN

Full Text Citing References

AN 1982:148618 CAPLUS
 DN 96:148618
 ED Entered STN: 12 May 1984
 TI The impact of coal mining on river ecology
 AU Edwards, R. W.
 CS Univ. Wales Inst. Sci. Technol., Cardiff, UK
 SO Min. Water Pollut., Inst. Water Eng. Sci. Sect. Symp. (1981), 3/1-3/8
 Publisher: Inst. Water Eng. Sci., London, UK.
 CODEN: 47KDAQ
 DT Conference; General Review
 LA English
 CC 61-0 (Water)
 Section cross-reference(s): 51
 AB A review with 28 refs.
 ST review river ecol coal mining
 IT Rivers
 (ecol. of, coal mining in relation to)
 IT Water pollution
 (river, by coal mining, ecol. in relation to)

IT Mines and Mining
 (coal, river ecol. in relation to)
 IT Ecology
 (river, coal mining in relation to)

L3 ANSWER 8 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN

Citing
 Text References

AN 1979:496139 CAPLUS
 DN 91:96139
 ED Entered STN: 12 May 1984
 TI Microbial ecology studies at two coal mine refuse sites in Illinois
 AU Miller, R. M.; Cameron, R. E.
 CS Argonne Natl. Lab., Argonne, IL, USA
 SO Report (1978), ANL/LRP-3, 48 pp. Avail.: NTIS
 From: Energy Res. Abstr. 1979, 4(5), Abstr. No. 9491
 DT Report
 LA English
 CC 60-2 (Sewage and Wastes)
 Section cross-reference(s): 51
 AB An investigation was made of the microflora assocd. with coal refuse at 2 abandoned mines in the midwestern US. Information was gathered for both the edaphic and the biotic compn. of the refuse material. Emphasis was placed on heterotrophic and autotrophic components as to nos., kinds, and physiol. groups. The presence of chemolithotrophs was also investigated. The relation between abiotic and biotic components in regard to distribution of bacteria, fungi, and algae is discussed. Information presented in this report will be utilized in assessing trends and changes in microbial nos. and compn. related to manipulations of the edaphic and biotic ecosystem components assocd. with reclamation of the refuse piles.
 ST microbial **ecol coal** mine tailing
 IT Microorganism
 (ecol. of, in coal mining refuse)
 IT Mining
 (of coal, refuse from, microbial ecol. in)
 IT Waste solids
 (coal tailings, reclamation of)

L3 ANSWER 9 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN

Citing
 Text References

AN 1975:509440 CAPLUS
 DN 83:109440
 ED Entered STN: 12 May 1984
 TI Lichen accumulation of some heavy metals from acidic surface substrates of coal mine ecosystems in southeastern Ohio
 AU Lawrey, James D.; Rudolph, Emanuel D.
 CS Dep. Bot., Ohio State Univ., Columbus, OH, USA
 SO Ohio Journal of Science (1975), 75(3), 113-17
 CODEN: OJSCA9; ISSN: 0030-0950
 DT Journal
 LA English
 CC 4-3 (Toxicology)
 AB Lichen samples from coal strip mines contained heavy metals at many times the concns. found in their substrates. Cladonia cristatella samples washed with water contained significantly less metals than untreated samples. Lichen material extd. with acetone showed no drop in metal content compared to untreated lichens. The accumulating nature of lichens in strip mine ecosystems makes them potential research tools in mineral cycling studies.
 ST heavy metal lichen strip mine; **ecol coal** mine lichen
 IT Mines
 (coal, heavy metals in lichens from)
 IT Cladonia

IT (heavy metals of, from strip mines)
 IT Ecology
 (of heavy metals, near coal strip mines, lichens in relation to)
 IT 7439-89-6, biological studies 7440-50-8, biological studies
 RL: BIOL (Biological study)
 (of lichens, from coal strip mines)
 IT 7429-90-5, biological studies 7439-95-4, biological studies 7439-96-5,
 biological studies 7439-98-7, biological studies 7440-09-7, biological
 studies 7440-66-6, biological studies 7440-70-2, biological studies
7723-14-0, biological studies
 RL: BIOL (Biological study)
 (of lichens, from strip mines)

L3 ANSWER 10 OF 10 CAPLUS COPYRIGHT 2004 ACS on STN

	Full Text	Citing References
AN	1975:93887 CAPLUS	
DN	82:93887	
ED	Entered STN: 12 May 1984	
TI	Toxicity of acid coal-mine spoils to plants	
AU	Berg, William A.; Vogel, Willis G.	
CS	Forest Serv., U. S. Dep. Agric., Berea, KY, USA	
SO	Ecol. Reclam. Devastated Land, [Proc. NATO Adv. Study Inst.] (1973), Meeting Date 1969, Volume 1, 57-68. Editor(s): Hutnik, Russell J.; Davis, Grant. Publisher: Gordon and Breach, New York, N. Y. CODEN: 29SNAK	
DT	Conference	
LA	English	
CC	4-3 (Toxicology)	
AB	Herbaceous legumes, shrub lespedezas, and black locust grown from seed in extremely acid spoils in the greenhouse and field, showed chlorosis on the margins of the leaves, which is indicative of Mn [7439-96-5] toxicity. Symptoms of Al [7429-90-5] toxicity were stubby roots without laterals. Soil pH was useful in predicting Mn toxicity to the legumes, but water-sol. Mn extd. from the spoils was not. Toxicity of extremely acid coal-mine spoils to plants was caused by excess sol. Mn and other metals, most probably Al. One yr after extremely acid spoils were mulched with hardwood chip, the pH of the top 30 cm was raised, while the total sol. salts and water-sol. Al were reduced.	
ST	coal mine spoil toxicity plants; ecol coal mine spoil plant; manganese mine spoil plant; aluminum mine spoil plant	
IT	Mines (coal, acid spoil of, toxicity of, to plants, aluminum and manganese in relation to)	
IT	Ecology (environmental damage, from coal-mine spoils)	
IT	<u>7429-90-5</u> , biological studies <u>7439-96-5</u> , biological studies RL: BIOL (Biological study) (of coal-mine spoils, toxicity of, to plants)	

```
=> s power coal
  468154 POWER
  21166 POWERS
  480788 POWER
    (POWER OR POWERS)
  209512 COAL
  35522 COALS
  211340 COAL
    (COAL OR COALS)
L4      858 POWER COAL
    (POWER(W) COAL)
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```
=> s caking coal
```

h eb c g cg b cg

eb

6210 CAKING
 3 CAKINGS
 6212 CAKING
 (CAKING OR CAKINGS)
 209512 COAL
 35522 COALS
 211340 COAL
 (COAL OR COALS)
 L5 1012 CAKING COAL
 (CAKING (W) COAL)

=> s 14 and 15
 L6 3 L4 AND L5

=> d 16 1-3 all

L6 ANSWER 1 OF 3 CAPLUS COPYRIGHT 2004 ACS on STN

Full Citing
 Text References

AN 1980:475344 CAPLUS
 DN 93:75344
 ED Entered STN: 12 May 1984
 TI Changes in the properties of concentrates obtained from different ranks of Karaganda coals in relation to separation densities
 AU Muzychuk, V. D.; Mel'nichuk, A. Yu.; Pluzhnikov, A. I.; Dobrovinskii, G. B.; Turchenkova, L. M.
 CS Karagand, Politekh. Inst., Karaganda, USSR
 SO Khimiya Tverdogo Topliva (Moscow, Russian Federation) (1980), (1), 10-16
 CODEN: KVTBTY; ISSN: 0023-1177
 DT Journal
 LA Russian
 CC 51-21 (Fossil Fuels, Derivatives, and Related Products)
 AB Beneficiation technols. for various coals depend on the properties of their concs. produced at different sepn. densities and are related to the quality of coke produced from them; concs. of certain Karaganda coals obtained at sepn. densities as low as 1800 kg/m³ have high caking power, and beneficiation technol. for these coals provides for their max. extn. into the coking conc.; fine grades of weakly **caking coals** are beneficiated at sepn. densities ≤1400kg/m³ in the prodn. of concs. of increased caking power and suitable for coking.
 ST coal conc sepn density; coking **power coal** beneficiation; coking coal caking power
 IT Carbonization and Coking
 (coals for, beneficiation of, caking properties in relation to, of Karaganda)
 IT Coal
 RL: USES (Uses)
 (coking, beneficiation of, caking properties in relation to, of Karaganda)

L6 ANSWER 2 OF 3 CAPLUS COPYRIGHT 2004 ACS on STN

Full Citing
 Text References

AN 1979:577805 CAPLUS
 DN 91:177805
 ED Entered STN: 12 May 1984
 TI Combustion Engineering low-Btu coal gasification process
 AU Richards, C. L.
 CS Combust. Eng., Inc., Windsor, CT, USA
 SO Report (1978), TIS-5862, CONF-780384-1, 8 pp. Avail.: Combustion Eng. Inc., Windsor, Conn
 From: Energy Res. Abstr. 1979, 4(11), Abstr. No. 28751
 DT Report
 LA English

h eb c g cg b cg

eb

CC 51-26 (Fossil Fuels, Derivatives, and Related Products)
 Section cross-reference(s): 47, 48

AB A design for air-blown entrainment gasification of coal at 1 atm. for elec. power generation permits use of most coals without special pretreatment (for **caking coals**), permits prodn. of a slagged ash, to minimize ash disposal problems, and consumes ≈100% of the C of the coal in the gasifier.

ST coal gasification low BTU; elec **power coal** gasification; design coal gasification; ash slagging coal gasification

IT Slags
 (ash removal by, in atm. coal gasification)

IT Power
 (generation of, low-BTU coal gasification in relation to)

IT Ashes (residues)
 (coal, removal of, in atm. gasification by slagging)

IT Fuel gas manufacturing
 (gasification, of coal to low-BTU gas)

L6 ANSWER 3 OF 3 CAPLUS COPYRIGHT 2004 ACS on STN

Full Text	Citing References
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AN 1968:106689 CAPLUS

DN 68:106689

ED Entered STN: 12 May 1984

TI Chemical characteristics of chloroform extracts from coking coal and their effects on caking power

AU Ihnatowicz, Maria; Lesniewski, Kazimierz

SO Prace Głównego Instytutu Górnictwa (1966), No. 393, 1-12
 CODEN: PGIGAT; ISSN: 0369-934X

DT Journal

LA Czech

CC 52 (Coal and Coal Derivatives)

AB Properties of CHCl₃ exts. from raw **ortho-caking coal** and also from **ortho-caking coals** and from vitrains sepd. from them and previously preheated to a detd. optimum temp. of 415° were examd. The CHCl₃ exts. obtained were divided into the following fractions: sol. in Et₂O, sol. and insol. in 10% H₂SO₄, in 10% H₂OH, and in n-hexane. Next, the fractions were sepd. in chromatographic columns. The properties of the CHCl₃ exts. examd. and their fractions were characterized by their elemental anal. and ir absorption spectra. Caking power of individual exts. and of their fractions were also detd. by adding various amts. to coals previously extd. or raw, and for mixts. thus obtained the caking power was detd. according to Roga. Depending on the method of prep. the coal (raw or preheated) and on the temp. of extn. (at the b.p. of CHCl₃ or at room temp.), individual CHCl₃ exts. differ from each other in the yield, elemental compn., caking power, and proportion and properties of fractions obtained from them by the use of solvents. The CHCl₃ exts. examd. contained 3 fractions: insol. in Et₂O and insol. and sol. in n-hexane; the first two had a higher caking power and were more aromatic than the other fraction. The tests carried out show that although the CHCl₃ exts. of **ortho-caking coals** markedly affect the caking power of coal, their specific part in the phenomenon of coal caking only takes place when they are organically linked with the whole coal system. 42 references.

ST COAL EXTS CAKING POWER; CHLOROFORM EXTS COAL; VITRAINS EXTS; COKING COAL;
CAKING POWER COAL EXTS

IT Coal
 RL: USES (Uses)
 (extract (chloroform) of, chem. characteristics of and caking power in relation to)

=> s 14 or 15

L7 1867 L4 OR L5

h ebc g cg b cg

eb

=> s 17 and **smokeless**
 2552 SMOKELESS
 L8 30 L7 AND SMOKELESS

=> s 18 and (**byproducts or by-products**)
 23414 BYPRODUCTS
 1246970 BY-PRODUCTS
 (PRODUCTS)
 L9 9 L8 AND (BYPRODUCTS OR BY-PRODUCTS)

=> d 19 1-9 all

L9 ANSWER 1 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN

	Full	Citing
	Text	References
AN	1964:468321	CAPLUS
DN	61:68321	
OREF	61:11809c-e	
ED	Entered STN: 22 Apr 2001	
TI	Pilot-plant studies on low-temperature carbonization of Indian coals. II. Products	
AU	Rao, K. Seshagiri; Rao, Y. V. Subba; Rao, D. K.; Agrawal, D. P.; Rao, B. S. Narayana; Vaidyeswaran, R.	
CS	Regional Res. Lab., Hyderabad	
SO	Low-Temp. Carbonization Non-Caking Coals Lignites Briquettes Coal Fines, Symp., Hyderabad, India, 1961 (1963), 1, 424-31	
DT	Journal	
LA	Unavailable	
CC	26 (Coal and Coal Derivatives)	
AB	The semicoke can be used as smokeless domestic fuel in place of charcoal. About 75-85% of the potential heat in the raw coal can be recovered in the form of semicoke, tar, and excess gas after meeting the heat requirements of the process. A min. temp. of 600° is necessary to assure the production of a semicoke that will not produce smoke when burnt on an open grate. The lighting time for a semicoke produced at 550° was 30% less than that of a semicoke produced at 650°. The 650° semicoke had 1-in. shatter indexes of 64-84%. The S contents of the heavy and light tars were 0.27-0.38 wt. %. Distn. of the tars up to 355° produced 29-95.6 vol. % distillates and 28-40 vol. % tar acids in the distillates. The aq. liquors produced contained 4.77-13.61 g./l. free NH ₃ , 1.87-5.49 g./l. fixed NH ₃ , and 4.97-6.91 g./l. tar acids. The high yield of tar acids is attributed to the highly oxygenated nature of the coals. The calorific value of the excess gas ranged from 96-148 B.t.u./cu. ft. due to inert gas contents of 77.3-85.5 vol. %.	
IT	Gas, fuel (manufactured) (by carbonization of noncaking and weakly caking coal by Lurgi-Spuelgas process)	
IT	Tar (low-temp., from noncaking and weakly caking coal carbonized by Lurgi-Spuelgas process)	
IT	Coke (semi- or low-temp., from noncaking and weakly caking coal)	

L9 ANSWER 2 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN

	Full	Citing
	Text	References
AN	1964:468320	CAPLUS
DN	61:68320	
OREF	61:11809a-c	
ED	Entered STN: 22 Apr 2001	
TI	Pilot-plant studies on low-temperature carbonization of Indian coals. I.	

AU Lurgi-Spuelgas process and operational data
 AU Rangrez, K. G.; Krishna, M. G.; Chowdhury, G. S.; Zaheer, S. H.
 CS Regional Res. Lab., Hyderabad
 SO Low-Temp. Carbonization Non-Caking Coals Lignites Briquettes. Coal Fines, Symp., Hyderabad, India, 1961 (1963), 1, 411-23
 DT Journal
 LA Unavailable
 CC 26 (Coal and Coal Derivatives)
 AB Data on the low-temp. carbonization of 6 noncaking and weakly **caking** coals are presented. A clean **smokeless** fuel for open-grates was produced at 650-750° in the Lurgi-Spuelgas-type carbonizer. The semicoke burned in an open-grate without any smoke or odor and was strong enough to withstand handling and transportation. The studies established the economic feasibility of large-scale low-temp. carbonization in India. For each ton of feed coal, the gas consumption varied from 1153 to 1274 cu. m. The pilot plant, based on the Lurgi-Spuelgas interval-heating system, has a daily capacity of 25 tons of 30-80 mm. coal. The main plant comprises a vertical shaft having a drying zone at the top, a carbonizing zone in the middle, and a semi coke cooling zone at the bottom; and a condensation system consisting of a precooler, a tar separator, and a pipe cooler. The integral parts of the carbonizer can withstand temps. ≥1000°. At carbonization temps of 640-750°, the yield of **products** in wt. % for the 6 coals studied were: semicoke 62.3-73.0, heavy tar 3.20-4.53, and light tar 2.02-5.36. Surplus gas varied from 87-188 m.3/ton of coal feed.
 IT Carbonization
 (low-temp. and/or semi-, of noncaking and weakly **caking** coal by Lurgi-Spuelgas process)

L9 ANSWER 3 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN

Full Text	Citing References
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AN 1935:49203 CAPLUS
 DN 29:49203
 OREF 29:6392h-i, 6393a-c
 ED Entered STN: 16 Dec 2001
 TI Recent experiments at the Fuels Research Station upon production of solid **smokeless** fuel
 AU Shaw, J. Fraser; King, J. G.
 SO Gas Journal (1934), 206;207, 779-82;39
 CODEN: GASJAF; ISSN: 0016-4941
 DT Journal
 LA Unavailable
 CC 21 (Fuels, Gas, Tar, and Coke)
 AB The retort of the Fuel Research Station is described. Rate of heat transfer through the coal at the top of the retort decreases with increase of caking power. Decrease of size of coal is equiv. in temp. control to an increase in caking power. The change from medium caking nuts to 3/4-in. slack is approx. equiv. to change from medium to strongly caking nuts. Data on carbonization yields and properties of **products** obtained under various conditions are compared. Lump coke decreased in combustibility with increasing throughput. There is a general decrease of ease of ignition and combustibility as caking power increases. Carbonization yields from air dry (I) and wetted (II) coal are compared. Coke and tar yields are higher for I for both coals; gas amt., cal. value and therms are less for I for 1 coal and less for II for the other. Variation in gas and tar yields and compn. with type and size of coal is discussed. The greater the percentage tar yield in the retort is of the yield in the Gray-King assay, the lighter is the tar. For all tests breeze was 9-18%. The shatter tests are very uniform. Friability as shown by the tumbler test increases slightly as the caking power of the coal increases and as the size decreases. Fuel requirements were 19.5-28.0 therms per ton of coal; it is least with sized coal of low coking power and greatest with **caking coals** of small size. Preheating

air reduces fuel requirements 10%. Yields and properties of **products** from carbonization of Silkstone and High Delf coal in Woodall-Duckham chamber ovens are given. The throughput is less than in the F. R. B. retort, but the bulk density of the coke is higher and the strength of the coke is greater; with suitable blends the fuel is almost as combustible as that from the F. R. B. retort.

- IT Carbonization
- IT Tar
 - (from dry and from wetted coals)
- IT Coal
 - (in **smokeless**-fuel production)
- IT Gas, illuminating and fuel
 - (retorts, of Fuel Research Station)
- IT Fuels
 - (**smokeless**, manuf. of)

L9 ANSWER 4 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN

Full	Citing
Text	References

- AN 1935:49202 CAPLUS
- DN 29:49202
- OREF 29:6392h-i, 6393a-c
- ED Entered STN: 16 Dec 2001
- TI Recent experiments at the Fuels Research Station upon production of solid **smokeless** fuel
- AU Shaw, J. Fraser; King, J. G.
- SO Gas World (1934), 100;101, 614-17, 670-2;59
- CODEN: GAWOAG; ISSN: 0016-5026
- DT Journal
- LA Unavailable
- CC 21 (Fuels, Gas, Tar, and Coke)
- AB The retort of the Fuel Research Station is described. Rate of heat transfer through the coal at the top of the retort decreases with increase of caking power. Decrease of size of coal is equiv. in temp. control to an increase in caking power. The change from medium caking nuts to 3/4-in. slack is approx. equiv. to change from medium to strongly caking nuts. Data on carbonization yields and properties of **products** obtained under various conditions are compared. Lump coke decreased in combustibility with increasing throughput. There is a general decrease of ease of ignition and combustibility as caking power increases. Carbonization yields from air dry (I) and wetted (II) coal are compared. Coke and tar yields are higher for I for both coals; gas amt., cal. value and therms are less for I for 1 coal and less for II for the other. Variation in gas and tar yields and compn. with type and size of coal is discussed. The greater the percentage tar yield in the retort is of the yield in the Gray-King assay, the lighter is the tar. For all tests breeze was 9-18%. The shatter tests are very uniform. Friability as shown by the tumbler test increases slightly as the caking power of the coal increases and as the size decreases. Fuel requirements were 19.5-28.0 therms per ton of coal; it is least with sized coal of low coking power and greatest with **caking coals** of small size. Preheating air reduces fuel requirements 10%. Yields and properties of **products** from carbonization of Silkstone and High Delf coal in Woodall-Duckham chamber ovens are given. The throughput is less than in the F. R. B. retort, but the bulk density of the coke is higher and the strength of the coke is greater; with suitable blends the fuel is almost as combustible as that from the F. R. B. retort.

- IT Carbonization
- IT Tar
 - (from dry and from wetted coals)
- IT Coal
 - (in **smokeless**-fuel production)
- IT Gas, illuminating and fuel
 - (retorts, of Fuel Research Station)

IT Fuels
 (smokeless, manuf. of)

L9 ANSWER 5 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN

Full Citing
 Text References

AN 1935:49201 CAPLUS
 DN 29:49201
 OREF 29:6392h-i,6393a-c
 ED Entered STN: 16 Dec 2001
 TI Recent experiments at the Fuels Research Station upon production of solid smokeless fuel
 AU Shaw, J. Fraser; King, J. G.
 SO Inst. Gas Engrs. Communication (1934), No. 88, 46 pp.
 DT Journal
 LA Unavailable
 CC 21 (Fuels, Gas, Tar, and Coke)
 AB The retort of the Fuel Research Station is described. Rate of heat transfer through the coal at the top of the retort decreases with increase of caking power. Decrease of size of coal is equiv. in temp. control to an increase in caking power. The change from medium caking nuts to 3/4-in. slack is approx. equiv. to change from medium to strongly caking nuts. Data on carbonization yields and properties of products obtained under various conditions are compared. Lump coke decreased in combustibility with increasing throughput. There is a general decrease of ease of ignition and combustibility as caking power increases. Carbonization yields from air dry (I) and wetted (II) coal are compared. Coke and tar yields are higher for I for both coals; gas amt., cal. value and therms are less for I for 1 coal and less for II for the other. Variation in gas and tar yields and compn. with type and size of coal is discussed. The greater the percentage tar yield in the retort is of the yield in the Gray-King assay, the lighter is the tar. For all tests breeze was 9-18%. The shatter tests are very uniform. Friability as shown by the tumbler test increases slightly as the caking power of the coal increases and as the size decreases. Fuel requirements were 19.5-28.0 therms per ton of coal; it is least with sized coal of low coking power and greatest with caking coals of small size. Preheating air reduces fuel requirements 10%. Yields and properties of products from carbonization of Silkstone and High Delf coal in Woodall-Duckham chamber ovens are given. The throughput is less than in the F. R. B. retort, but the bulk density of the coke is higher and the strength of the coke is greater; with suitable blends the fuel is almost as combustible as that from the F. R. B. retort.
 IT Carbonization
 IT Tar
 (from dry and from wetted coals)
 IT Coal
 (in smokeless-fuel production)
 IT Gas, illuminating and fuel
 (retorts, of Fuel Research Station)
 IT Fuels
 (smokeless, manuf. of)

L9 ANSWER 6 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN

Full Citing
 Text References

AN 1923:23738 CAPLUS
 DN 17:23738
 OREF 17:3594i,3595a-b
 ED Entered STN: 16 Dec 2001
 TI Low-temperature carbonization in vertical retorts
 AU Anon.
 SO Gas Journal (1923), 163, 581-2
 CODEN: GASJAF; ISSN: 0016-4941

DT Journal
 LA Unavailable
 CC 21 (Fuels, Gas, Tar, and Coke)
 AB Tests were carried out on Glover-West vertical retorts at an av. temp. of 780° with a 60-40% mixt. of caking and non-caking coals with 0, 7.24, 13.47 and 20%, resp., of steam. The retorts were not adapted to low-temp. carbonization, yet the results showed that if a setting of similar construction with Fe retorts should be used, it should be possible successfully to manuf. a good smokeless fuel by this method of carbonization. The yields of by-products were as follows: with 0, 7.24, 13.47 and 20%, resp., the gas make was 7190, 6700, 7350 and 7750 cu. ft. per ton; gross calorific value 640, 671, 661, and 640 B.t.u., the gas differing from the usual low-temp. gas in having a much lower % of unsatd. hydrocarbons, nearly twice as much H, and a much lower amt. of satd. hydrocarbons; tar 12.72, 14.18, 15.22 and 16.62 gal.; (NH₄)₂SO₄ 21.4, 20.6, 18.05 and 28.2 lbs. The heat required varied from 6.3 therms with 0, to 5.3 therms with 7.24% steam and to 12.4 therms with 20% steam.
 IT Carbonization
 (low-temp., in vertical retorts)
 IT 7727-37-9, Nitrogen
 (in coal distn. products from vertical retorts, distribution of)
 IT 110-86-1, Pyridine
 (in coal-distn. products from vertical retorts)

L9 ANSWER 7 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN

	Full Text	Citing References
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AN 1923:23737 CAPLUS
 DN 17:23737
 OREF 17:3594i,3595a-b
 ED Entered STN: 16 Dec 2001
 TI Low-temperature carbonization in vertical retorts
 AU Anon.
 SO Gas World (1923), 79, 130-41
 CODEN: GAWOAG; ISSN: 0016-5026
 DT Journal
 LA Unavailable
 CC 21 (Fuels, Gas, Tar, and Coke)
 AB Tests were carried out on Glover-West vertical retorts at an av. temp. of 780° with a 60-40% mixt. of caking and non-caking coals with 0, 7.24, 13.47 and 20%, resp., of steam. The retorts were not adapted to low-temp. carbonization, yet the results showed that if a setting of similar construction with Fe retorts should be used, it should be possible successfully to manuf. a good smokeless fuel by this method of carbonization. The yields of by-products were as follows: with 0, 7.24, 13.47 and 20%, resp., the gas make was 7190, 6700, 7350 and 7750 cu. ft. per ton; gross calorific value 640, 671, 661, and 640 B.t.u., the gas differing from the usual low-temp. gas in having a much lower % of unsatd. hydrocarbons, nearly twice as much H, and a much lower amt. of satd. hydrocarbons; tar 12.72, 14.18, 15.22 and 16.62 gal.; (NH₄)₂SO₄ 21.4, 20.6, 18.05 and 28.2 lbs. The heat required varied from 6.3 therms with 0, to 5.3 therms with 7.24% steam and to 12.4 therms with 20% steam.
 IT Carbonization
 (low-temp., in vertical retorts)
 IT 7727-37-9, Nitrogen
 (in coal distn. products from vertical retorts, distribution of)
 IT 110-86-1, Pyridine
 (in coal-distn. products from vertical retorts)

L9 ANSWER 8 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN

	Full Text	Citing References
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h ebc g cg b cg

eb

AN 1923:23736 CAPLUS
 DN 17:23736
 OREF 17:3594i,3595a-b
 ED Entered STN: 16 Dec 2001
 TI Low-temperature carbonization in vertical retorts
 AU Anon.
 SO Fuel Research Board, Tech. Paper (1923), No. 7,
 DT Journal
 LA Unavailable
 CC 21 (Fuels, Gas, Tar, and Coke)
 AB Tests were carried out on Glover-West vertical retorts at an av. temp. of 780° with a 60-40% mixt. of caking and non-caking coals with 0, 7.24, 13.47 and 20%, resp., of steam. The retorts were not adapted to low-temp. carbonization, yet the results showed that if a setting of similar construction with Fe retorts should be used, it should be possible successfully to manuf. a good smokeless fuel by this method of carbonization. The yields of by-products were as follows: with 0, 7.24, 13.47 and 20%, resp., the gas make was 7190, 6700, 7350 and 7750 cu. ft. per ton; gross calorific value 640, 671, 661, and 640 B.t.u., the gas differing from the usual low-temp. gas in having a much lower % of unsatd. hydrocarbons, nearly twice as much H, and a much lower amt. of satd. hydrocarbons; tar 12.72, 14.18, 15.22 and 16.62 gal.; (NH₄)₂SO₄ 21.4, 20.6, 18.05 and 28.2 lbs. The heat required varied from 6.3 therms with 0, to 5.3 therms with 7.24% steam and to 12.4 therms with 20% steam.
 IT Carbonization
 (low-temp., in vertical retorts)
 IT 7727-37-9, Nitrogen
 (in coal distn. products from vertical retorts, distribution of)
 IT 110-86-1, Pyridine
 (in coal-distn. products from vertical retorts)

L9 ANSWER 9 OF 9 CAPLUS COPYRIGHT 2004 ACS on STN

Full	Citing
Text	References

AN 1919:6152 CAPLUS
 DN 13:6152
 OREF 13:1141h-i,1142a-f
 ED Entered STN: 16 Dec 2001
 TI Low-temperature carbonization in relation to the production of motor spirit, fuel oils, smokeless fuel and power gas: its aims and objectives
 AU Marshall, F. D.
 SO Gas Journal (1919), 145, 383-5,451-4
 CODEN: GASJAF; ISSN: 0016-4941
 DT Journal
 LA Unavailable
 CC 21 (Fuels, Gas, Tar, and Coke)
 AB M. proposes to reduce the abnormal waste in utilizing fuel by subjecting it to the following cycle of operations: (1) Carbonization of the raw coal at a low temp. and recovery of the liquid products and a portion of the NH₃; (2) gasification of the resultant coke into producer gas with recovery of the bulk of the NH₃; each ton of coke will yield 120,000-140,000 cu. ft. of gas with a calorific value of 120-130 B. t. u. per cu. ft.; (3) conversion of the power gas into electricity; 110 cu. ft. of 130 B. t. u. gas will develop in good engines 1 kw. hr., and if for raising steam, 200 cu. ft. will develop 1 kw. hr. in efficient steam turbines. An av. coal when submitted to a high temp. in the retorts of 2000-2200° F. will yield approx. per ton: gas 12,000-13,000 cu. ft. of approx. 500 B. t. u., coke 66%, tar 9-10 gals., (NH₄)₂SO₄ 20-28 lbs.; to a low temp. of 900-1200° F.: gas 4000-6000 cu. ft. of approx. 650 B. t. u., coke 70-75%, tar oils 18-22 gals., sulfate 15-22 lbs. Solid, compact and transportable low-temp. coke can be produced by confining the coal in comparatively narrow chambers with walls sufficiently strong to withstand the pressure of expansion of the coke.

Under the low temp., a swelling coke will reach its max. of expansion in about 3.5 hrs. and will then shrink to allow of easy discharge. Such coke will also contain a residue of 9-12% of volatile matter which makes it easily ignitable and satisfactory for fuel. Its calorific value compared with good coal is 13,300 B. t. u., coal being 14,200 B. t. u. Low-temp. gas varies in thermal value from 550 to 650 B. t. u. before washing, after washing 400-450 B. t. u. Low-temp. carbonization of an ordinary bituminous coal gave results per ton as follows: Coke 75% containing 9% inflammable gas, very hard and dense; crude oils, water free 22 gals, sp. gr. 1.06; ammoniacal liquor, 9.5%; motor spirit, 3.94 gals.; pitch, 23%; ash in coke, 10.3%. Cannel coal gave results as follows: Coke, 70%, poor for domestic purposes but good for producer gas; ash in coke, 20.4%; crude oils, 53.5 gals.; ammoniacal liquor, 0.5%; pitch, 29%; motor spirit, 8.75 gals. The process described is the Tozer process, originally devized by the Tarless Fuel Syndicate. It consists in distributing the coal in thin layers in the form of a circle in such a way that a charge of 20-25 cwt. of coal can be carbonized in 4.5 hrs. The retort is constructed with 2 annular sections, one disposed concentrically within the other, the exterior heat being conducted through the first annulus of coal to the second by ribs which divide the annuli into vertical cells, advantage being taken of the conductivity of the cast iron of which the retort is built. The inner tube formed by the surrounding annuli is not filled with coal, but acts as a gas passage connecting the upper and lower ends of the retort. The retort will deal with practically every kind of carbonaceous material, caking and non-caking coals, shales, lignites and peat, in every physical condition from 3-in. down to dust, and will carbonize and produce coke from colliery slack and washings. The retorts are heated by a specially designed recuperative setting.

IT Gas
 (illuminating and fuel, from low-temp. carbonization)
 IT Gas
 (illuminating and fuel, gasoline detn. in)
 IT Fuels
 (internal-combustion, from low-temp. carbonization)
 IT Carbonization
 (low-temp.)

=> s ca2313882/pn
 L10 0 CA2313882/PN

=> s ca2313882/pa
 L11 0 CA2313882/PA

=> s ca2313882
 L12 0 CA2313882

=> s goraczko
 0 GORACZKO
 L13 0 GORACZKO

=> s valorization
 281 VALORIZATION
 2 VALORIZATIONS
 L14 283 VALORIZATION
 (VALUEZATION OR VALORIZATIONS)

=> s l14 and coal
 209512 COAL
 35522 COALS
 211340 COAL
 (COAL OR COALS)
 L15 17 L14 AND COAL

=> d 115 1-17 ti

L15 ANSWER 1 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

**Citing
References**

TI Life cycle assessment of **coal** by-products based electric power production scenarios

L15 ANSWER 2 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

**Citing
References**

TI Use of wastes in glasses, glass-ceramics and ceramics

L15 ANSWER 3 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

**Citing
References**

TI **Valorization** of textile sludges

L15 ANSWER 4 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

**Citing
References**

TI Rheology and pipeline transport of combustible pulps

L15 ANSWER 5 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

**Citing
References**

TI Optimization of combustion of **coal-washery** products which are difficult for **valorization**

L15 ANSWER 6 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

**Citing
References**

TI Quality control and industrial reutilization potential of FGD gypsum

L15 ANSWER 7 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

**Citing
References**

TI Upgrading of waste derived solid fuel by steam gasification

L15 ANSWER 8 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

**Citing
References**

TI Scrap tire pyrolysis: Are the effluents valuable products?

L15 ANSWER 9 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

**Citing
References**

TI The **valorization** of carbon containing wastes in classical coking plants

L15 ANSWER 10 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

**Citing
References**

TI Ultrasonic removal of heavy metals and pyrolytic **valorization** of waste oils

L15 ANSWER 11 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

**Citing
References**

TI Thermal byproducts treatment and **valorization**

L15 ANSWER 12 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

**Citing
References**

TI Multinutrient fertilizer

L15 ANSWER 13 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

Citing
 References

TI **Valorization** of waste mining products by vitrification

L15 ANSWER 14 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

Citing
 References

TI Aspects of the chemical **valorization** of **coal**

L15 ANSWER 15 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

Citing
 References

TI Aspects of the chemical **valorization** of **coal** gases in Belgium

L15 ANSWER 16 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

Citing
 References

TI Aspects of the chemical **valorization** of **coal** gases in Belgium

L15 ANSWER 17 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

Citing
 References

TI **Valorization** of mined fuels

=> dl15 13-17 all

DL15 IS NOT A RECOGNIZED COMMAND

The previous command name entered was not recognized by the system.
For a list of commands available to you in the current file, enter
"HELP COMMANDS" at an arrow prompt (>).

=> d 115 13-17 all

L15 ANSWER 13 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

Full
 Text Citing
 References

AN	1971:452305	CAPLUS
DN	75:52305	
ED	Entered STN:	12 May 1984
TI	Valorization of waste mining products by vitrification	
AU	Santt, Rene	
CS	Fr.	
SO	Annales des Mines (1971), (March), 43-56	
	CODEN: ANMSA3; ISSN: 0003-4282	
DT	Journal	
LA	French	
CC	57 (Ceramics)	
AB	Wastes from coal and iron mines can be saved by vitrification. The resulting product is glass. This black glass turns opaque during the vitrification process because of the Fe content Proper thermal treatment of this glass causes small crystals to form in the mass giving a vitreous ceramic. Glass designation 195 has the best velocity of recrystn. and it resists thermal shock very well. The compn. of this glass is SiO ₂ 42.25, Al ₂ O ₃ 5.84, Fe ₂ O ₃ 17.64, CaO 24.04, MgO 0.59, MnO 0.15, Na ₂ O 5.91, and CaF ₂ 0.68%.	
ST	glass coal iron mine waste; coal mine waste glass; iron mine waste glass	
IT	Glass ceramics	
	Glass	
	RL: USES (Uses)	
	(opaque, from mine wastes)	
IT	Mines	
	(wastes from, black glass from)	
IT	<u>1309-37-1</u> , uses and miscellaneous <u>1344-43-0</u> <u>7789-75-5</u> , uses and	

miscellaneous
 RL: USES (Uses)
 (glass, opaque, from mine wastes)

L15 ANSWER 14 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

Full Citing
 Text References

AN 1950:25919 CAPLUS
 DN 44:25919
 OREF 44:5077g
 ED Entered STN: 22 Apr 2001
 TI Aspects of the chemical **valorization** of **coal**
 AU Ferrero, P.
 SO Exp. nat. ind. chim. Charleroi (1945) 58-63
 DT Journal
 LA Unavailable
 CC 21 (Fuels and Carbonization Products)
 AB cf. C.A. 43, 8118i. The potentialities of **coal** as a source of raw materials are described.
 IT Coal
 (as chemical source)
 IT Chemicals
 (from **coal**, evaluation of)

L15 ANSWER 15 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

Full Citing
 Text References

AN 1949:45112 CAPLUS
 DN 43:45112
 OREF 43:8118i,8119a
 ED Entered STN: 22 Apr 2001
 TI Aspects of the chemical **valorization** of **coal** gases in Belgium
 AU Ferrero, Paul
 SO Gas Journal (1949), 259, 105-6
 CODEN: GASJAF; ISSN: 0016-4941
 DT Journal
 LA Unavailable
 CC 21 (Fuels and Carbonization Products)
 AB H was first recovered from **coal** gas for use in the synthesis of NH₃. CH₄ is used in part for the prepn. of make-up H or for synthesis gas to be converted into CH₃OH. C₂H₄ is useful for the synthesis of a no. of chemicals.
 IT Fuel gas
 (as chemical raw material)
 IT 74-82-8, Methane
 (recovery from **coal** gas and use of)
 IT 74-85-1, Ethylene
 (recovery of, from **coal** gas)
 IT 1333-74-0, Hydrogen
 (recovery of, from coke-oven gas)

L15 ANSWER 16 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

Full Citing
 Text References

AN 1949:45111 CAPLUS
 DN 43:45111
 OREF 43:8118i,8119a
 ED Entered STN: 22 Apr 2001
 TI Aspects of the chemical **valorization** of **coal** gases in Belgium
 AU Ferrero, Paul
 SO Gas World (1949), 130, 1066
 CODEN: GAWOAG; ISSN: 0016-5026
 DT Journal
 LA Unavailable

CC 21 (Fuels and Carbonization Products)
 AB H was first recovered from **coal** gas for use in the synthesis of NH₃.
 CH₄ is used in part for the prepn. of make-up H or for synthesis gas to be converted into CH₃OH. C₂H₄ is useful for the synthesis of a no. of chemicals.
 IT Fuel gas
 (as chemical raw material)
 IT 74-82-8, Methane
 (recovery from **coal** gas and use of)
 IT 74-85-1, Ethylene
 (recovery of, from **coal** gas)
 IT 1333-74-0, Hydrogen
 (recovery of, from coke-oven gas)

L15 ANSWER 17 OF 17 CAPLUS COPYRIGHT 2004 ACS on STN

Full Citing
 Text References

AN 1928:39699 CAPLUS
 DN 22:39699
 OREF 22:4758f-g
 ED Entered STN: 16 Dec 2001
 TI **Valorization** of mined fuels
 AU Folliet, A.
 SO Revue de Chimie Industrielle et le Moniteur Scientifique de Quesneville
 Reunis (Paris) (1928), 37, 182-6
 CODEN: RCHJAR
 DT Journal
 LA Unavailable
 CC 21 (Fuels, Gas, Tar, and Coke)
 AB A review of the different furnaces used in the low-temp. distn. of **coal**.
 IT Furnace
 (distn., for **coal** at low-temps.)
 IT Fuels
 (liquid)
 IT Carbonization
 (low-temp.)
 IT Fuels
 (**valorization** of mined)

=> log y

	SINCE FILE	TOTAL
	ENTRY	SESSION
COST IN U.S. DOLLARS		
FULL ESTIMATED COST	121.06	191.63
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE	TOTAL
CA SUBSCRIBER PRICE	ENTRY	SESSION
	-18.20	-31.50

STN INTERNATIONAL LOGOFF AT 08:55:18 ON 14 SEP 2004